

# Cadina-1(6),4-diene

<b>Inchi:</b>	InChI=1S/C15H24/c1-10(2)13-8-6-12(4)14-7-5-11(3)9-15(13)14/h10-11H,5-9H2,1-4H3
<b>InchiKey:</b>	KBMTZMXTAFFCCM-UHFFFAOYSA-N
<b>Formula:</b>	C15H24
<b>SMILES:</b>	CC1=C2CCC(C)CC2=C(C(C)C)CC1
<b>Mol. weight [g/mol]:</b>	204.35

## Physical Properties

Property code	Value	Unit	Source
gf	175.19	kJ/mol	Joback Method
hf	-147.23	kJ/mol	Joback Method
hfus	18.77	kJ/mol	Joback Method
hvap	52.65	kJ/mol	Joback Method
log10ws	-5.11		Crippen Method
logp	4.869		Crippen Method
mcvol	191.890	ml/mol	McGowan Method
pc	1984.12	kPa	Joback Method
rinpol	1457.00		NIST Webbook
rinpol	1442.00		NIST Webbook
rinpol	1457.00		NIST Webbook
rinpol	1463.00		NIST Webbook
ripol	1712.00		NIST Webbook
ripol	1657.00		NIST Webbook
ripol	1712.00		NIST Webbook
tb	595.63	K	Joback Method
tc	813.11	K	Joback Method
tf	321.45	K	Joback Method
vc	0.725	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	502.77	J/molxK	595.63	Joback Method
cpg	523.96	J/molxK	631.88	Joback Method
cpg	543.93	J/molxK	668.12	Joback Method

cpg	562.72	J/molxK	704.37	Joback Method
cpg	580.40	J/molxK	740.62	Joback Method
cpg	597.01	J/molxK	776.86	Joback Method
cpg	612.61	J/molxK	813.11	Joback Method
dvisc	0.0020246	Paxs	321.45	Joback Method
dvisc	0.0011622	Paxs	367.15	Joback Method
dvisc	0.0007544	Paxs	412.84	Joback Method
dvisc	0.0005337	Paxs	458.54	Joback Method
dvisc	0.0004020	Paxs	504.24	Joback Method
dvisc	0.0003174	Paxs	549.93	Joback Method
dvisc	0.0002599	Paxs	595.63	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R287644&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R287644&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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