

2-(Ethyl(m-tolyl)amino)ethyl 2,3,4,5,6-pentafluorobenzoate

Inchi:	InChI=1S/C18H16F5NO2/c1-3-24(11-6-4-5-10(2)9-11)7-8-26-18(25)12-13(19)15(21)17(2
InchiKey:	MBIZDDGNLXPDRM-UHFFFAOYSA-N
Formula:	C18H16F5NO2
SMILES:	CCN(CCOC(=O)c1c(F)c(F)c(F)c(F)c1F)c1cccc(C)c1
Mol. weight [g/mol]:	373.32

Physical Properties

Property code	Value	Unit	Source
gf	-829.47	kJ/mol	Joback Method
hf	-1168.43	kJ/mol	Joback Method
hfus	49.33	kJ/mol	Joback Method
hvap	71.30	kJ/mol	Joback Method
log10ws	-5.82		Crippen Method
logp	4.374		Crippen Method
mcvol	243.230	ml/mol	McGowan Method
pc	1531.86	kPa	Joback Method
rinqol	2069.00		NIST Webbook
tb	779.56	K	Joback Method
tc	972.47	K	Joback Method
tf	528.16	K	Joback Method
vc	0.960	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	700.11	J/molxK	779.56	Joback Method
cpg	713.28	J/molxK	811.71	Joback Method
cpg	725.58	J/molxK	843.86	Joback Method
cpg	737.02	J/molxK	876.02	Joback Method
cpg	747.64	J/molxK	908.17	Joback Method
cpg	757.44	J/molxK	940.32	Joback Method
cpg	766.46	J/molxK	972.47	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U373602&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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