

# 4-Penten-2-ol, bromoacetate

<b>Inchi:</b>	InChI=1S/C7H11BrO2/c1-3-4-6(2)10-7(9)5-8/h3,6H,1,4-5H2,2H3
<b>InchiKey:</b>	CJFYOQXFISOMFE-UHFFFAOYSA-N
<b>Formula:</b>	C7H11BrO2
<b>SMILES:</b>	C=CCC(C)OC(=O)CBr
<b>Mol. weight [g/mol]:</b>	207.06

## Physical Properties

Property code	Value	Unit	Source
gf	-126.14	kJ/mol	Joback Method
hf	-286.13	kJ/mol	Joback Method
hfus	17.16	kJ/mol	Joback Method
hvap	45.71	kJ/mol	Joback Method
log10ws	-2.01		Crippen Method
logp	1.889		Crippen Method
mcvol	130.130	ml/mol	McGowan Method
pc	3360.64	kPa	Joback Method
rinpola	1076.00		NIST Webbook
ripola	1575.00		NIST Webbook
tb	498.25	K	Joback Method
tc	698.62	K	Joback Method
tf	283.85	K	Joback Method
vc	0.488	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	259.41	J/molxK	498.25	Joback Method
cpg	269.90	J/molxK	531.65	Joback Method
cpg	279.89	J/molxK	565.04	Joback Method
cpg	289.37	J/molxK	598.44	Joback Method
cpg	298.37	J/molxK	631.83	Joback Method
cpg	306.90	J/molxK	665.23	Joback Method
cpg	314.96	J/molxK	698.62	Joback Method
dvisc	0.0032347	Paxs	283.85	Joback Method

dvisc	0.0017256	Paxs	319.58	Joback Method
dvisc	0.0010445	Paxs	355.32	Joback Method
dvisc	0.0006930	Paxs	391.05	Joback Method
dvisc	0.0004925	Paxs	426.78	Joback Method
dvisc	0.0003690	Paxs	462.52	Joback Method
dvisc	0.0002881	Paxs	498.25	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R26532&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R26532&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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