

# 3-Pentanone, 2-methyl-

<b>Other names:</b>	2-METHYL-3-PENTANONE 2-Methyl-3-pentanal 2-Methylpentan-3-one 4-Methyl-3-pentanone Ethyl isopropyl ketone Isopropyl ethyl ketone iso-C <sub>3</sub> H <sub>7</sub> COC <sub>2</sub> H <sub>5</sub>
<b>Inchi:</b>	InChI=1S/C6H12O/c1-4-6(7)5(2)3/h5H,4H2,1-3H3
<b>InchiKey:</b>	HYTRYEXINDDXJK-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>6</sub> H <sub>12</sub> O
<b>SMILES:</b>	CCC(=O)C(C)C
<b>Mol. weight [g/mol]:</b>	100.16
<b>CAS:</b>	565-69-5

## Physical Properties

Property code	Value	Unit	Source
chl	-3750.16 ± 0.88	kJ/mol	NIST Webbook
gf	-131.72	kJ/mol	Joback Method
hf	-286.10 ± 0.90	kJ/mol	NIST Webbook
hfl	-325.90 ± 0.90	kJ/mol	NIST Webbook
hfus	9.37	kJ/mol	Joback Method
hvap	39.80 ± 0.20	kJ/mol	NIST Webbook
hvap	40.50	kJ/mol	NIST Webbook
hvap	39.79 ± 0.22	kJ/mol	NIST Webbook
ie	9.10 ± 0.01	eV	NIST Webbook
ie	9.10 ± 0.01	eV	NIST Webbook
log10ws	-1.37		Crippen Method
logp	1.621		Crippen Method
mcvol	96.970	ml/mol	McGowan Method
pc	3368.44	kPa	Joback Method
rinpol	734.00		NIST Webbook
rinpol	732.00		NIST Webbook
rinpol	733.00		NIST Webbook
rinpol	733.00		NIST Webbook
rinpol	722.00		NIST Webbook
rinpol	732.00		NIST Webbook
rinpol	733.00		NIST Webbook

rinpol	733.00		NIST Webbook
rinpol	748.00		NIST Webbook
rinpol	737.00		NIST Webbook
rinpol	722.00		NIST Webbook
rinpol	726.00		NIST Webbook
rinpol	747.90		NIST Webbook
rinpol	722.00		NIST Webbook
rinpol	732.00		NIST Webbook
rinpol	732.00		NIST Webbook
rinpol	737.00		NIST Webbook
rinpol	752.00		NIST Webbook
rinpol	742.00		NIST Webbook
rinpol	742.00		NIST Webbook
rinpol	733.00		NIST Webbook
rinpol	732.00		NIST Webbook
rinpol	732.00		NIST Webbook
ripol	1000.00		NIST Webbook
ripol	997.00		NIST Webbook
ripol	1007.00		NIST Webbook
ripol	1003.00		NIST Webbook
ripol	972.00		NIST Webbook
ripol	989.00		NIST Webbook
ripol	1005.00		NIST Webbook
ripol	1003.00		NIST Webbook
ripol	1003.00		NIST Webbook
ripol	997.00		NIST Webbook
ripol	1003.00		NIST Webbook
tb	386.15 ± 2.00	K	NIST Webbook
tb	387.25 ± 1.00	K	NIST Webbook
tb	387.15 ± 1.00	K	NIST Webbook
tb	388.15 ± 1.00	K	NIST Webbook
tb	388.20 ± 1.00	K	NIST Webbook
tb	386.20	K	NIST Webbook
tb	387.70 ± 1.00	K	NIST Webbook
tb	386.15 ± 2.00	K	NIST Webbook
tb	386.00 ± 0.50	K	NIST Webbook
tb	387.65 ± 1.00	K	NIST Webbook
tb	388.15 ± 1.00	K	NIST Webbook
tb	389.00 ± 2.00	K	NIST Webbook
tb	387.90 ± 1.00	K	NIST Webbook
tb	387.15 ± 2.00	K	NIST Webbook
tb	390.15 ± 3.00	K	NIST Webbook
tb	390.65 ± 5.00	K	NIST Webbook
tb	368.15	K	NIST Webbook

tb	386.90 ± 1.00	K	NIST Webbook
tb	384.15 ± 3.00	K	NIST Webbook
tb	386.00 ± 3.00	K	NIST Webbook
tb	390.15 ± 1.00	K	NIST Webbook
tc	571.85	K	Joback Method
tf	192.31	K	Joback Method
vc	0.371	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	225.77	J/mol×K	541.56	Joback Method
cpg	207.72	J/mol×K	480.98	Joback Method
cpg	198.11	J/mol×K	450.69	Joback Method
cpg	188.09	J/mol×K	420.40	Joback Method
cpg	177.67	J/mol×K	390.11	Joback Method
cpg	216.93	J/mol×K	511.27	Joback Method
cpg	234.22	J/mol×K	571.85	Joback Method
dvisc	0.0061867	Paxs	192.31	Joback Method
dvisc	0.0002910	Paxs	390.11	Joback Method
dvisc	0.0003828	Paxs	357.14	Joback Method
dvisc	0.0005326	Paxs	324.18	Joback Method
dvisc	0.0007984	Paxs	291.21	Joback Method
dvisc	0.0013273	Paxs	258.24	Joback Method
dvisc	0.0025603	Paxs	225.28	Joback Method
hvapt	36.20	kJ/mol	413.50	NIST Webbook
hvapt	43.40	kJ/mol	343.50	NIST Webbook
hvapt	41.00	kJ/mol	333.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.48028e+01
Coeff. B	-3.45648e+03
Coeff. C	-4.86110e+01
Temperature range (K), min.	286.74

Temperature range (K), max.	412.79
-----------------------------	--------

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	2.16937e+02
Coeff. B	-1.31300e+04
Coeff. C	-3.05877e+01
Coeff. D	2.58098e-05
Temperature range (K), min.	200.00
Temperature range (K), max.	567.00

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>KDB:</b>	<a href="https://www.thermo.com/files/research/kdb/mol/mol1203.mol">https://www.thermo.com/files/research/kdb/mol/mol1203.mol</a>
<b>Crippen Method:</b>	<a href="https://www.chemed.com/doc/models/crippen_log10ws">https://www.chemed.com/doc/models/crippen_log10ws</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C565695&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C565695&amp;Units=SI</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1203">https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1203</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Air-Water Partitioning of C5 and C6 Alkanones: Measurement, Critical Comparison, and Correlation</b>	<a href="https://www.doi.org/10.1021/acs.jced.9b00726">https://www.doi.org/10.1021/acs.jced.9b00726</a>
<b>The Yaws Handbook of Vapor Pressure</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Recommended Data:</b>	
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/35-782-5/3-Pentanone-2-methyl.pdf>

Generated by Cheméo on 2024-04-23 15:15:50.950572345 +0000 UTC m=+16174599.871149666.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.