

Propanedinitrile, (phenylmethylene)-

Other names:	Malononitrile, benzylidene- «alpha»-Cyanocinnamionitrile «beta», «beta»-Dicyanostyrene Benzalmalononitrile Benzylidenemalonodinitrile Benzylidenemalononitrile 2-Phenyl-1,1-dicyanoethylene 2,2-Dicyano-1-phenylethylene Benzalmalonitrile BMN Benzal-malonitril F 2370 1,1-Dicyano-2-phenylethene 1,1-Dicyano-2-phenylethylene 2-Cyano-3-phenylacrylonitrile 2-Phenyl-1,1-dicyanoethene Malonodinitrile, benzylidene Benzylidenemalonitrile NSC 490 Propanedinitrile, 2-(phenylmethylene)- «beta», «beta»-Styrenedicarbonitrile beta,beta-styrenedicarbonitrile
Inchi:	InChI=1S/C10H6N2/c11-7-10(8-12)6-9-4-2-1-3-5-9/h1-6H
InchiKey:	WAVNYPVYNSIHNC-UHFFFAOYSA-N
Formula:	C10H6N2
SMILES:	<chem>N#CC(C#N)=Cc1ccccc1</chem>
Mol. weight [g/mol]:	154.17
CAS:	2700-22-3

Physical Properties

Property code	Value	Unit	Source
gf	483.76	kJ/mol	Joback Method
hf	423.99	kJ/mol	Joback Method
hfus	17.60	kJ/mol	Joback Method
hvap	61.12	kJ/mol	Joback Method
log10ws	-2.87		Crippen Method

logp	2.117		Crippen Method
mcvol	126.460	ml/mol	McGowan Method
pc	2953.69	kPa	Joback Method
rinpol	1473.00		NIST Webbook
rinpol	1473.00		NIST Webbook
rinpol	1487.00		NIST Webbook
tb	663.08	K	Joback Method
tc	914.90	K	Joback Method
tf	339.82	K	Joback Method
vc	0.520	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	283.38	J/mol×K	663.08	Joback Method
cpg	292.43	J/mol×K	705.05	Joback Method
cpg	300.72	J/mol×K	747.02	Joback Method
cpg	308.32	J/mol×K	788.99	Joback Method
cpg	315.31	J/mol×K	830.96	Joback Method
cpg	321.77	J/mol×K	872.93	Joback Method
cpg	327.76	J/mol×K	914.90	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C2700223&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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