

# Benzene, 1-methoxy-3-nitro-

<b>Other names:</b>	Anisole, m-nitro- m-Methoxynitrobenzene m-Nitroanisole 1-Methoxy-3-nitrobenzene 3-Methoxynitrobenzene 3-Nitroanisole Methyl m-nitrophenyl ether NSC 4956
<b>Inchi:</b>	InChI=1S/C7H7NO3/c1-11-7-4-2-3-6(5-7)8(9)10/h2-5H,1H3
<b>InchiKey:</b>	WGYFINWERLNPFR-UHFFFAOYSA-N
<b>Formula:</b>	C7H7NO3
<b>SMILES:</b>	<chem>COc1cccc([N+](=O)[O-])c1</chem>
<b>Mol. weight [g/mol]:</b>	153.14
<b>CAS:</b>	555-03-3

## Physical Properties

Property code	Value	Unit	Source
ea	1.04 ± 0.10	eV	NIST Webbook
gf	41.39	kJ/mol	Joback Method
hf	-105.73	kJ/mol	Joback Method
hfus	20.09	kJ/mol	Joback Method
hvap	53.12	kJ/mol	Joback Method
ie	9.10 ± 0.10	eV	NIST Webbook
ie	9.01	eV	NIST Webbook
ie	8.70	eV	NIST Webbook
log10ws	-2.24		Crippen Method
logp	1.603		Crippen Method
mcvol	109.020	ml/mol	McGowan Method
pc	4010.84	kPa	Joback Method
rinpol	228.00		NIST Webbook
rinpol	1346.10		NIST Webbook
rinpol	228.00		NIST Webbook
rinpol	1346.10		NIST Webbook
rinpol	1273.00		NIST Webbook
tb	531.20	K	NIST Webbook
tc	811.60	K	Joback Method
tf	311.15 ± 1.50	K	NIST Webbook

vc

0.419

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	242.84	J/mol×K	565.48	Joback Method
cpg	253.81	J/mol×K	606.50	Joback Method
cpg	264.03	J/mol×K	647.52	Joback Method
cpg	273.53	J/mol×K	688.54	Joback Method
cpg	282.32	J/mol×K	729.56	Joback Method
cpg	290.41	J/mol×K	770.58	Joback Method
cpg	297.83	J/mol×K	811.60	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	395.20	K	1.00	NIST Webbook

## Sources

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C555033&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

## Legend

**cpg:** Ideal gas heat capacity

**ea:** Electron affinity

**gf:** Standard Gibbs free energy of formation

**hf:** Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/35-823-9/Benzene-1-methoxy-3-nitro.pdf>

Generated by Cheméo on 2024-04-26 09:25:35.319343305 +0000 UTC m=+16412784.239920617.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.