

# trans-Bicyclo[3.3.0]octane

**Inchi:** InChI=1S/C8H14/c1-3-7-5-2-6-8(7)4-1/h7-8H,1-6H2/t7-,8-  
**InchiKey:** AEBWATHAIVJLTA-ZKCHVHJHSA-N  
**Formula:** C8H14  
**SMILES:** C1CC2CCCC2C1  
**Mol. weight [g/mol]:** 110.20  
**CAS:** 5597-89-7

## Physical Properties

Property code	Value	Unit	Source
chl	-5040.00 ± 2.00	kJ/mol	NIST Webbook
chl	-5041.70 ± 5.00	kJ/mol	NIST Webbook
gf	113.78	kJ/mol	Joback Method
hf	-67.00 ± 3.00	kJ/mol	NIST Webbook
hfl	-109.00 ± 2.00	kJ/mol	NIST Webbook
hfus	8.55	kJ/mol	Joback Method
hvap	42.70 ± 0.80	kJ/mol	NIST Webbook
hvap	42.00	kJ/mol	NIST Webbook
hvap	42.70 ± 0.80	kJ/mol	NIST Webbook
log10ws	-2.48		Crippen Method
logp	2.587		Crippen Method
mvol	101.860	ml/mol	McGowan Method
pc	3560.02	kPa	Joback Method
tb	407.00 ± 3.00	K	NIST Webbook
tb	405.00 ± 3.00	K	NIST Webbook
tb	409.40 ± 2.00	K	NIST Webbook
tb	405.00 ± 2.00	K	NIST Webbook
tc	614.38	K	Joback Method
tf	237.00 ± 3.00	K	NIST Webbook
tf	244.00 ± 3.00	K	NIST Webbook
tf	243.00 ± 3.00	K	NIST Webbook
vc	0.382	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	198.70	J/mol×K	404.46	Joback Method
cpg	217.06	J/mol×K	439.45	Joback Method
cpg	234.29	J/mol×K	474.43	Joback Method
cpg	250.43	J/mol×K	509.42	Joback Method
cpg	265.55	J/mol×K	544.41	Joback Method
cpg	279.71	J/mol×K	579.39	Joback Method
cpg	292.95	J/mol×K	614.38	Joback Method
cpl	180.30	J/mol×K	308.00	NIST Webbook
dvisc	0.0013767	Paxs	208.76	Joback Method
dvisc	0.0010230	Paxs	241.38	Joback Method
dvisc	0.0008158	Paxs	273.99	Joback Method
dvisc	0.0006827	Paxs	306.61	Joback Method
dvisc	0.0005912	Paxs	339.23	Joback Method
dvisc	0.0005251	Paxs	371.84	Joback Method
dvisc	0.0004754	Paxs	404.46	Joback Method
hvapt	41.40	kJ/mol	309.00	NIST Webbook
hvapt	41.30 ± 0.40	kJ/mol	320.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	405.20	K	101.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5597897&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5597897&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

**chl:** Standard liquid enthalpy of combustion

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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