

# Dimethylmalonic acid, heptadecyl 2-phenethyl ester

Inchi:	InChI=1S/C30H50O4/c1-4-5-6-7-8-9-10-11-12-13-14-15-16-17-21-25-33-28(31)30(2,3)29
InchiKey:	VIKULCUMGIVBOS-UHFFFAOYSA-N
Formula:	C30H50O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C(C)(C)C(=O)OCCc1ccccc1
Mol. weight [g/mol]:	474.72

## Physical Properties

Property code	Value	Unit	Source
gf	-150.87	kJ/mol	Joback Method
hf	-924.35	kJ/mol	Joback Method
hfus	65.66	kJ/mol	Joback Method
hvap	101.67	kJ/mol	Joback Method
log10ws	-8.97		Crippen Method
logp	8.213		Crippen Method
mcvol	424.680	ml/mol	McGowan Method
pc	748.15	kPa	Joback Method
rinsol	3239.00		NIST Webbook
tb	1061.83	K	Joback Method
tc	1308.19	K	Joback Method
tf	601.02	K	Joback Method
vc	1.645	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1497.51	J/molxK	1061.83	Joback Method
cpg	1516.73	J/molxK	1102.89	Joback Method
cpg	1534.24	J/molxK	1143.95	Joback Method
cpg	1550.17	J/molxK	1185.01	Joback Method
cpg	1564.65	J/molxK	1226.07	Joback Method
cpg	1577.81	J/molxK	1267.13	Joback Method
cpg	1589.78	J/molxK	1308.19	Joback Method
dvisc	0.0001763	Paxs	601.02	Joback Method
dvisc	0.0000807	Paxs	677.82	Joback Method

dvisc	0.0000433	Paxs	754.62	Joback Method
dvisc	0.0000261	Paxs	831.42	Joback Method
dvisc	0.0000171	Paxs	908.23	Joback Method
dvisc	0.0000120	Paxs	985.03	Joback Method
dvisc	0.0000088	Paxs	1061.83	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361627&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361627&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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