

# Silvex, methyl ester

<b>Other names:</b>	Propanoic acid, 2-(2,4,5-trichlorophenoxy)-, methyl ester Propionic acid, 2-(2,4,5-trichlorophenoxy)-, methyl ester Methyl 2-(2,4,5-trichlorophenoxy)propionate 2-(2,4,5-Trichlorophenoxy)propionic acid methyl ester 2,4,5-TP, methyl ester Fenoprop, methyl ester Silvex, methylated
<b>Inchi:</b>	InChI=1S/C10H9Cl3O3/c1-5(10(14)15-2)16-9-4-7(12)6(11)3-8(9)13/h3-5H,1-2H3
<b>InchiKey:</b>	YTAXYXOJOYIQQO-UHFFFAOYSA-N
<b>Formula:</b>	C10H9Cl3O3
<b>SMILES:</b>	<chem>COC(=O)C(C)Oc1cc(Cl)c(Cl)cc1Cl</chem>
<b>Mol. weight [g/mol]:</b>	283.54
<b>CAS:</b>	4841-20-7

## Physical Properties

Property code	Value	Unit	Source
gf	-260.31	kJ/mol	Joback Method
hf	-477.13	kJ/mol	Joback Method
hfus	27.57	kJ/mol	Joback Method
hvap	66.45	kJ/mol	Joback Method
log10ws	-3.88		Crippen Method
logp	3.587		Crippen Method
mcvol	178.030	ml/mol	McGowan Method
pc	2587.22	kPa	Joback Method
rinpol	1709.00		NIST Webbook
rinpol	1709.00		NIST Webbook
rinpol	1705.00		NIST Webbook
rinpol	1705.00		NIST Webbook
rinpol	1712.00		NIST Webbook
rinpol	1716.00		NIST Webbook
rinpol	1733.00		NIST Webbook
ripol	2371.00		NIST Webbook
ripol	2372.00		NIST Webbook
ripol	2371.00		NIST Webbook
tb	680.38	K	Joback Method
tc	908.89	K	Joback Method
tf	361.16 ± 0.20	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	393.73	J/molxK	680.38	Joback Method
cpg	404.23	J/molxK	718.47	Joback Method
cpg	414.01	J/molxK	756.55	Joback Method
cpg	423.08	J/molxK	794.64	Joback Method
cpg	431.40	J/molxK	832.72	Joback Method
cpg	438.98	J/molxK	870.81	Joback Method
cpg	445.80	J/molxK	908.89	Joback Method
dvisc	0.0007966	Paxs	435.59	Joback Method
dvisc	0.0005196	Paxs	476.39	Joback Method
dvisc	0.0003625	Paxs	517.19	Joback Method
dvisc	0.0002666	Paxs	557.99	Joback Method
dvisc	0.0002044	Paxs	598.78	Joback Method
dvisc	0.0001622	Paxs	639.58	Joback Method
dvisc	0.0001323	Paxs	680.38	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4841207&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4841207&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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