

# 1H,3H-Naphtho[1,8-cd]pyran-1-one

<b>Other names:</b>	Peri-Naphthalide Naphthalid Naphthalide 1-Naphthalenecarboxylic acid, 8-(hydroxymethyl)-, «delta»-lactone 1-Naphthoic acid, 8-(hydroxymethyl)-, «delta»-lactone 1,8-Naphthalide
<b>Inchi:</b>	InChI=1S/C12H8O2/c13-12-10-6-2-4-8-3-1-5-9(7-14-12)11(8)10/h1-6H,7H2
<b>InchiKey:</b>	UKOVZLWSUZKTRL-UHFFFAOYSA-N
<b>Formula:</b>	C12H8O2
<b>SMILES:</b>	O=C1OCc2cccc3cccc1c23
<b>Mol. weight [g/mol]:</b>	184.19
<b>CAS:</b>	518-86-5

## Physical Properties

Property code	Value	Unit	Source
gf	109.71	kJ/mol	Joback Method
hf	-62.91	kJ/mol	Joback Method
hfus	21.67	kJ/mol	Joback Method
hvap	56.52	kJ/mol	Joback Method
log10ws	-3.87		Crippen Method
logp	2.510		Crippen Method
mcvol	133.300	ml/mol	McGowan Method
pc	3740.80	kPa	Joback Method
rinpol	1828.00		NIST Webbook
tb	635.76	K	Joback Method
tc	896.34	K	Joback Method
tf	426.13	K	Joback Method
vc	0.507	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	330.53	J/molxK	635.76	Joback Method
cpg	343.79	J/molxK	679.19	Joback Method

cpg	355.97	J/mol×K	722.62	Joback Method
cpg	367.18	J/mol×K	766.05	Joback Method
cpg	377.51	J/mol×K	809.48	Joback Method
cpg	387.04	J/mol×K	852.91	Joback Method
cpg	395.89	J/mol×K	896.34	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C518865&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C518865&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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