

# 4'-(2-Bromoethyl)benzoic acid

<b>Other names:</b>	p-(«beta»-Bromoethyl)benzoic acid p-(b-Bromoethyl)benzoic acid Benzoic acid, 4-(2-bromoethyl)- 4-(2-bromoethyl)benzoic acid
<b>Inchi:</b>	InChI=1S/C9H9BrO2/c10-6-5-7-1-3-8(4-2-7)9(11)12/h1-4H,5-6H2,(H,11,12)
<b>InchiKey:</b>	BKMRWJWLBBHHGCF-UHFFFAOYSA-N
<b>Formula:</b>	C9H9BrO2
<b>SMILES:</b>	O=C(O)c1ccc(CCBrc1)
<b>Mol. weight [g/mol]:</b>	229.07
<b>CAS:</b>	52062-92-7

## Physical Properties

Property code	Value	Unit	Source
gf	-123.74	kJ/mol	Joback Method
hf	-242.51	kJ/mol	Joback Method
hfus	23.69	kJ/mol	Joback Method
hvap	68.43	kJ/mol	Joback Method
log10ws	-2.77		Crippen Method
logp	2.322		Crippen Method
mcvol	138.850	ml/mol	McGowan Method
pc	4205.63	kPa	Joback Method
tb	649.19	K	Joback Method
tc	865.25	K	Joback Method
tf	400.68	K	Joback Method
vc	0.518	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	312.02	J/molxK	649.19	Joback Method
cpg	352.08	J/molxK	829.24	Joback Method
cpg	345.18	J/molxK	793.23	Joback Method
cpg	337.77	J/molxK	757.22	Joback Method
cpg	329.79	J/molxK	721.21	Joback Method

cpg	321.22	J/mol×K	685.20	Joback Method
cpg	358.50	J/mol×K	865.25	Joback Method
dvisc	0.0000851	Paxs	649.19	Joback Method
dvisc	0.0001214	Paxs	607.77	Joback Method
dvisc	0.0001826	Paxs	566.35	Joback Method
dvisc	0.0002928	Paxs	524.94	Joback Method
dvisc	0.0005092	Paxs	483.52	Joback Method
dvisc	0.0009823	Paxs	442.10	Joback Method
dvisc	0.0021704	Paxs	400.68	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C52062927&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C52062927&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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