

# Pyrazine, 1,4-dioxide

<b>Other names:</b>	Pyrazine di-N-oxide Pyrazine dioxide Pyrazine N,N'-dioxide
<b>Inchi:</b>	InChI=1S/C4H4N2O2/c7-5-1-2-6(8)4-3-5/h1-4H
<b>InchiKey:</b>	SXTKIFFXFIDYJF-UHFFFAOYSA-N
<b>Formula:</b>	C4H4N2O2
<b>SMILES:</b>	[O-][n+]1cc[n+](O-)cc1
<b>Mol. weight [g/mol]:</b>	112.09
<b>CAS:</b>	2423-84-9

## Physical Properties

Property code	Value	Unit	Source
chs	-2215.30 ± 1.60	kJ/mol	NIST Webbook
hf	186.50 ± 1.90	kJ/mol	NIST Webbook
hfs	116.90 ± 0.80	kJ/mol	NIST Webbook
hsub	69.60 ± 1.70	kJ/mol	NIST Webbook
hsub	116.90 ± 0.80	kJ/mol	NIST Webbook
ie	8.33 ± 0.02	eV	NIST Webbook
log10ws	-4.53		Crippen Method
logp	-1.047		Crippen Method
mcvol	75.160	ml/mol	McGowan Method
tf	574.00 ± 4.00	K	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2423849&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2423849&amp;Units=SI</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>tf:</b>	Normal melting (fusion) point

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