

# (E)-1-(3-Nitrophenyl)ethanone oxime

<b>Other names:</b>	1-(3-Nitrophenyl)-1-ethanone oxime
<b>Inchi:</b>	InChI=1S/C8H8N2O3/c1-6(9-11)7-3-2-4-8(5-7)10(12)13/h2-5,11H,1H3
<b>InchiKey:</b>	PDRKZXUNOXCETR-UHFFFAOYSA-N
<b>Formula:</b>	C8H8N2O3
<b>SMILES:</b>	CC(=NO)c1cccc([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	180.16
<b>CAS:</b>	7471-32-1

## Physical Properties

Property code	Value	Unit	Source
hf	-73.95	kJ/mol	Joback Method
hvap	73.00	kJ/mol	Joback Method
log10ws	-1.84		Crippen Method
logp	1.793		Crippen Method
mcvol	128.790	ml/mol	McGowan Method
pc	3602.88	kPa	Joback Method
rinpol	1807.00		NIST Webbook
tb	734.68	K	Joback Method
tc	975.28	K	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7471321&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7471321&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

hf: Enthalpy of formation at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature

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