

cis-1,1-Difluoro-2-methyl-3-vinylcyclopropane

Inchi:	InChI=1S/C6H8F2/c1-3-5-4(2)6(5,7)8/h3-5H,1H2,2H3
InchiKey:	AHHUGNMVDWMJFL-UHFFFAOYSA-N
Formula:	C6H8F2
SMILES:	C=CC1C(C)C1(F)F
Mol. weight [g/mol]:	118.12
CAS:	79517-49-0

Physical Properties

Property code	Value	Unit	Source
gf	-262.30	kJ/mol	Joback Method
hf	-386.60	kJ/mol	Joback Method
hfus	10.15	kJ/mol	Joback Method
hvap	24.79	kJ/mol	Joback Method
log10ws	-1.91		Crippen Method
logp	2.074		Crippen Method
mcvol	83.780	ml/mol	McGowan Method
pc	3329.68	kPa	Joback Method
tb	329.54	K	Joback Method
tc	497.76	K	Joback Method
tf	190.16	K	Joback Method
vc	0.342	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	143.76	J/molxK	329.54	Joback Method
cpg	155.87	J/molxK	357.58	Joback Method
cpg	167.14	J/molxK	385.61	Joback Method
cpg	177.62	J/molxK	413.65	Joback Method
cpg	187.36	J/molxK	441.69	Joback Method
cpg	196.42	J/molxK	469.73	Joback Method
cpg	204.87	J/molxK	497.76	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C79517490&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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