

2,2,2-Trichloroethanol, heptafluorobutyrate

Inchi:	InChI=1S/C6H2Cl3F7O2/c7-3(8,9)1-18-2(17)4(10,11)5(12,13)6(14,15)16/h1H2
InchiKey:	BPRJWOQYSBUXBU-UHFFFAOYSA-N
Formula:	C6H2Cl3F7O2
SMILES:	O=C(OCC(Cl)(Cl)Cl)C(F)(F)C(F)(F)C(F)(F)F
Mol. weight [g/mol]:	345.43

Physical Properties

Property code	Value	Unit	Source
gf	-1622.38	kJ/mol	Joback Method
hf	-1866.96	kJ/mol	Joback Method
hfus	18.58	kJ/mol	Joback Method
hvap	40.36	kJ/mol	Joback Method
log10ws	-4.05		Crippen Method
logp	3.733		Crippen Method
mvol	151.950	ml/mol	McGowan Method
pc	2271.90	kPa	Joback Method
rinpol	902.00		NIST Webbook
tb	507.23	K	Joback Method
tc	683.24	K	Joback Method
tf	333.11	K	Joback Method
vc	0.625	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	345.11	J/mol×K	507.23	Joback Method
cpg	353.72	J/mol×K	536.57	Joback Method
cpg	361.52	J/mol×K	565.90	Joback Method
cpg	368.55	J/mol×K	595.24	Joback Method
cpg	374.89	J/mol×K	624.57	Joback Method
cpg	380.56	J/mol×K	653.91	Joback Method
cpg	385.64	J/mol×K	683.24	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U375579&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvac:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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