

# Glutaric acid, di(4-methoxyphenyl) ester

<b>Inchi:</b>	InChI=1S/C19H20O6/c1-22-14-6-10-16(11-7-14)24-18(20)4-3-5-19(21)25-17-12-8-15(23
<b>InchiKey:</b>	URNPJOHPEQQDBO-UHFFFAOYSA-N
<b>Formula:</b>	C19H20O6
<b>SMILES:</b>	COc1ccc(OC(=O)CCCC(=O)Oc2ccc(OC)cc2)cc1
<b>Mol. weight [g/mol]:</b>	344.36

## Physical Properties

Property code	Value	Unit	Source
gf	-363.18	kJ/mol	Joback Method
hf	-739.41	kJ/mol	Joback Method
hfus	40.22	kJ/mol	Joback Method
hvap	86.90	kJ/mol	Joback Method
log10ws	-4.41		Crippen Method
logp	3.385		Crippen Method
mcvol	257.670	ml/mol	McGowan Method
pc	1807.70	kPa	Joback Method
rinpol	2947.00		NIST Webbook
rinpol	2947.00		NIST Webbook
tb	894.86	K	Joback Method
tc	1119.29	K	Joback Method
tf	570.55	K	Joback Method
vc	0.968	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	776.84	J/molxK	894.86	Joback Method
cpg	825.13	J/molxK	1081.89	Joback Method
cpg	818.39	J/molxK	1044.48	Joback Method
cpg	810.18	J/molxK	1007.08	Joback Method
cpg	800.52	J/molxK	969.67	Joback Method
cpg	789.40	J/molxK	932.27	Joback Method
cpg	830.40	J/molxK	1119.29	Joback Method
dvisc	0.0000370	Paxs	894.86	Joback Method

dvisc	0.0000460	Paxs	840.81	Joback Method
dvisc	0.0000590	Paxs	786.76	Joback Method
dvisc	0.0000785	Paxs	732.71	Joback Method
dvisc	0.0001093	Paxs	678.65	Joback Method
dvisc	0.0001612	Paxs	624.60	Joback Method
dvisc	0.0002558	Paxs	570.55	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U358751&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U358751&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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