

# Phthalic acid, 3,3-dimethylbut-2-yl isohexyl ester

Inchi:	InChI=1S/C20H30O4/c1-14(2)10-9-13-23-18(21)16-11-7-8-12-17(16)19(22)24-15(3)20(4)
InchiKey:	ASXIHOZXNTWIPE-UHFFFAOYSA-N
Formula:	C20H30O4
SMILES:	CC(C)CCCOC(=O)c1ccccc1C(=O)OC(C)C(C)(C)C
Mol. weight [g/mol]:	334.45

## Physical Properties

Property code	Value	Unit	Source
gf	-249.58	kJ/mol	Joback Method
hf	-739.98	kJ/mol	Joback Method
hfus	32.32	kJ/mol	Joback Method
hvap	79.29	kJ/mol	Joback Method
log10ws	-5.77		Crippen Method
logp	4.871		Crippen Method
mcvol	283.780	ml/mol	McGowan Method
pc	1376.84	kPa	Joback Method
rinpol	2165.00		NIST Webbook
tb	837.13	K	Joback Method
tc	1046.77	K	Joback Method
tf	470.84	K	Joback Method
vc	1.073	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	879.79	J/molxK	837.13	Joback Method
cpg	951.11	J/molxK	1011.83	Joback Method
cpg	939.11	J/molxK	976.89	Joback Method
cpg	926.03	J/molxK	941.95	Joback Method
cpg	911.81	J/molxK	907.01	Joback Method
cpg	896.41	J/molxK	872.07	Joback Method
cpg	962.05	J/molxK	1046.77	Joback Method
dvisc	0.0000365	Paxs	837.13	Joback Method
dvisc	0.0000494	Paxs	776.08	Joback Method

dvisc	0.0000705	Paxs	715.03	Joback Method
dvisc	0.0001074	Paxs	653.99	Joback Method
dvisc	0.0001786	Paxs	592.94	Joback Method
dvisc	0.0003336	Paxs	531.89	Joback Method
dvisc	0.0007329	Paxs	470.84	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357004&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357004&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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