

# Butanoic acid, 4-chloro-3-oxo-, ethyl ester

<b>Other names:</b>	Ethyl 4-chloroacetoacetate Ethyl «gamma»-chloroacetoacetate 4-Chloroacetoacetic acid ethyl ester Acetoacetic acid, 4-chloro-, ethyl ester
<b>Inchi:</b>	InChI=1S/C6H9ClO3/c1-2-10-6(9)3-5(8)4-7/h2-4H2,1H3
<b>InchiKey:</b>	OHLRLMWUFVDREV-UHFFFAOYSA-N
<b>Formula:</b>	C6H9ClO3
<b>SMILES:</b>	CCOC(=O)CC(=O)CCl
<b>Mol. weight [g/mol]:</b>	164.59
<b>CAS:</b>	638-07-3

## Physical Properties

Property code	Value	Unit	Source
gf	-375.13	kJ/mol	Joback Method
hf	-540.29	kJ/mol	Joback Method
hfus	19.88	kJ/mol	Joback Method
hvap	49.24	kJ/mol	Joback Method
log10ws	-0.63		Crippen Method
logp	0.747		Crippen Method
mcvol	116.650	ml/mol	McGowan Method
pc	3372.36	kPa	Joback Method
tb	504.27	K	Joback Method
tc	698.73	K	Joback Method
tf	309.39	K	Joback Method
vc	0.451	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	238.35	J/molxK	504.27	Joback Method
cpg	247.32	J/molxK	536.68	Joback Method
cpg	255.91	J/molxK	569.09	Joback Method
cpg	264.12	J/molxK	601.50	Joback Method
cpg	271.94	J/molxK	633.91	Joback Method

cpg	279.39	J/molxK	666.32	Joback Method
cpg	286.45	J/molxK	698.73	Joback Method
dvisc	0.0026181	Paxs	309.39	Joback Method
dvisc	0.0015720	Paxs	341.87	Joback Method
dvisc	0.0010312	Paxs	374.35	Joback Method
dvisc	0.0007236	Paxs	406.83	Joback Method
dvisc	0.0005350	Paxs	439.31	Joback Method
dvisc	0.0004124	Paxs	471.79	Joback Method
dvisc	0.0003287	Paxs	504.27	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	388.20	K	1.90	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C638073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C638073&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure

<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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