

# Nonanoyl chloride

<b>Other names:</b>	Pelargonoyl chloride Nonanoic acid chloride
<b>Inchi:</b>	InChI=1S/C9H17ClO/c1-2-3-4-5-6-7-8-9(10)11/h2-8H2,1H3
<b>InchiKey:</b>	NTQYXUJLILNTFH-UHFFFAOYSA-N
<b>Formula:</b>	C9H17ClO
<b>SMILES:</b>	CCCCCCCCC(=O)Cl
<b>Mol. weight [g/mol]:</b>	176.68
<b>CAS:</b>	764-85-2

## Physical Properties

Property code	Value	Unit	Source
gf	-115.95	kJ/mol	Joback Method
hf	-357.41	kJ/mol	Joback Method
hfus	24.86	kJ/mol	Joback Method
hvap	46.76	kJ/mol	Joback Method
log10ws	-3.52		Crippen Method
logp	3.502		Crippen Method
mvol	151.480	ml/mol	McGowan Method
pc	2365.67	kPa	Joback Method
rinpol	1215.70		NIST Webbook
tb	493.15 ± 6.00	K	NIST Webbook
tb	487.00 ± 10.00	K	NIST Webbook
tb	488.50	K	NIST Webbook
tc	676.51	K	Joback Method
tf	271.04	K	Joback Method
vc	0.595	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	327.38	J/mol×K	496.62	Joback Method
cpg	340.41	J/mol×K	526.60	Joback Method
cpg	352.87	J/mol×K	556.58	Joback Method
cpg	364.77	J/mol×K	586.56	Joback Method

cpg	376.14	J/molxK	616.55	Joback Method
cpg	386.99	J/molxK	646.53	Joback Method
cpg	397.34	J/molxK	676.51	Joback Method
dvisc	0.0041228	Paxs	271.04	Joback Method
dvisc	0.0020309	Paxs	308.64	Joback Method
dvisc	0.0011667	Paxs	346.23	Joback Method
dvisc	0.0007472	Paxs	383.83	Joback Method
dvisc	0.0005181	Paxs	421.43	Joback Method
dvisc	0.0003814	Paxs	459.02	Joback Method
dvisc	0.0002942	Paxs	496.62	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	382.20	K	2.90	NIST Webbook
tbrp	371.20	K	2.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C764852&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C764852&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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