

# Propanoic acid, 2-methyl-, 3-phenylpropyl ester

Other names:

Hydrocinnamyl isobutyrate  
Isobutyric acid, 3-phenylpropyl ester  
3-Phenylpropyl isobutyrate  
Phenylpropyl iso-butyrate

Inchi:

InChI=1S/C13H18O2/c1-11(2)13(14)15-10-6-9-12-7-4-3-5-8-12/h3-5,7-8,11H,6,9-10H2,1

InchiKey:

VBTAKMZSMFMLGT-UHFFFAOYSA-N

Formula:

C13H18O2

SMILES:

CC(C)C(=O)OCCc1ccccc1

Mol. weight [g/mol]:

206.28

CAS:

103-58-2

## Physical Properties

Property code	Value	Unit	Source
gf	-65.37	kJ/mol	Joback Method
hf	-325.20	kJ/mol	Joback Method
hfus	22.73	kJ/mol	Joback Method
hvap	55.58	kJ/mol	Joback Method
log10ws	-2.99		Crippen Method
logp	2.818		Crippen Method
mcvol	177.710	ml/mol	McGowan Method
pc	2309.17	kPa	Joback Method
rinpol	1408.00		NIST Webbook
rinpol	1490.00		NIST Webbook
ripol	1996.00		NIST Webbook
tb	599.37	K	Joback Method
tc	806.35	K	Joback Method
tf	319.85	K	Joback Method
vc	0.673	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	443.71	J/mol×K	599.37	Joback Method
cpg	459.89	J/mol×K	633.87	Joback Method

cpg	475.13	J/mol×K	668.36	Joback Method
cpg	489.48	J/mol×K	702.86	Joback Method
cpg	502.95	J/mol×K	737.36	Joback Method
cpg	515.57	J/mol×K	771.85	Joback Method
cpg	527.37	J/mol×K	806.35	Joback Method
dvisc	0.0027865	Paxs	319.85	Joback Method
dvisc	0.0012714	Paxs	366.44	Joback Method
dvisc	0.0006925	Paxs	413.02	Joback Method
dvisc	0.0004266	Paxs	459.61	Joback Method
dvisc	0.0002873	Paxs	506.20	Joback Method
dvisc	0.0002068	Paxs	552.78	Joback Method
dvisc	0.0001567	Paxs	599.37	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C103582&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C103582&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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