

# Ethanone, 1-(4-ethylphenyl)-

<b>Other names:</b>	Acetophenone, 4'-ethyl- p-Acetylethylbenzene p-Ethylacetophenone p-Ethylphenyl methyl ketone 4'-Ethylacetophenone 4-Ethylacetophenone 1-(4-Ethylphenyl)ethanone
<b>Inchi:</b>	InChI=1S/C10H12O/c1-3-9-4-6-10(7-5-9)8(2)11/h4-7H,3H2,1-2H3
<b>InchiKey:</b>	NODGRWCMFMEGJH-UHFFFAOYSA-N
<b>Formula:</b>	C10H12O
<b>SMILES:</b>	CCc1ccc(C(C)=O)cc1
<b>Mol. weight [g/mol]:</b>	148.20
<b>CAS:</b>	937-30-4

## Physical Properties

Property code	Value	Unit	Source
gf	7.18	kJ/mol	Joback Method
hf	-137.25	kJ/mol	Joback Method
hfus	16.91	kJ/mol	Joback Method
hvap	47.54	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	2.452		Crippen Method
mcvol	129.570	ml/mol	McGowan Method
pc	3089.85	kPa	Joback Method
rinpol	1281.00		NIST Webbook
rinpol	1228.00		NIST Webbook
rinpol	1245.00		NIST Webbook
rinpol	1267.00		NIST Webbook
rinpol	1274.00		NIST Webbook
rinpol	1228.00		NIST Webbook
rinpol	1267.00		NIST Webbook
rinpol	1277.00		NIST Webbook
rinpol	1277.00		NIST Webbook
rinpol	1281.00		NIST Webbook
rinpol	1295.00		NIST Webbook
rinpol	1273.80		NIST Webbook
rinpol	1281.00		NIST Webbook

ripol	1867.00		NIST Webbook
ripol	1867.00		NIST Webbook
tb	512.00 ± 2.00	K	NIST Webbook
tc	729.82	K	Joback Method
tf	248.90 ± 0.25	K	NIST Webbook
tf	252.60 ± 2.00	K	NIST Webbook
vc	0.493	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	276.66	J/mol×K	513.73	Joback Method
cpg	290.22	J/mol×K	549.75	Joback Method
cpg	303.00	J/mol×K	585.76	Joback Method
cpg	315.02	J/mol×K	621.78	Joback Method
cpg	326.32	J/mol×K	657.79	Joback Method
cpg	336.91	J/mol×K	693.81	Joback Method
cpg	346.84	J/mol×K	729.82	Joback Method
dvisc	0.0012729	Paxs	328.40	Joback Method
dvisc	0.0022588	Paxs	291.33	Joback Method
dvisc	0.0008058	Paxs	365.46	Joback Method
dvisc	0.0005549	Paxs	402.53	Joback Method
dvisc	0.0004070	Paxs	439.60	Joback Method
dvisc	0.0003132	Paxs	476.66	Joback Method
dvisc	0.0002503	Paxs	513.73	Joback Method
hvapt	42.20	kJ/mol	331.00	NIST Webbook
hvapt	39.80	kJ/mol	358.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	398.20	K	2.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C937304&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C937304&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>d<sub>visc</sub>:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>h<sub>vapt</sub>:</b>	Enthalpy of vaporization at a given temperature
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ri<sub>npol</sub>:</b>	Non-polar retention indices
<b>ri<sub>pol</sub>:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tb<sub>rp</sub>:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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