

# Benzene, 4-fluoro-1-methyl-2-nitro-

<b>Other names:</b>	Toluene, 4-fluoro-2-nitro- 4-Fluoro-2-nitrotoluene
<b>Inchi:</b>	InChI=1S/C7H6FNO2/c1-5-2-3-6(8)4-7(5)9(10)11/h2-4H,1H3
<b>InchiKey:</b>	SKWTUNAAJNDEIK-UHFFFAOYSA-N
<b>Formula:</b>	C7H6FNO2
<b>SMILES:</b>	Cc1ccc(F)cc1[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	155.13
<b>CAS:</b>	446-10-6

## Physical Properties

Property code	Value	Unit	Source
gf	-58.05	kJ/mol	Joback Method
hf	-181.09	kJ/mol	Joback Method
hfus	21.59	kJ/mol	Joback Method
hvap	50.55	kJ/mol	Joback Method
log10ws	-2.93		Crippen Method
logp	2.042		Crippen Method
mvol	104.920	ml/mol	McGowan Method
pc	3801.00	kPa	Joback Method
tb	547.31	K	Joback Method
tc	785.21	K	Joback Method
tf	364.31	K	Joback Method
vc	0.419	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	228.86	J/molxK	547.31	Joback Method
cpg	239.05	J/molxK	586.96	Joback Method
cpg	248.55	J/molxK	626.61	Joback Method
cpg	257.38	J/molxK	666.26	Joback Method
cpg	265.58	J/molxK	705.91	Joback Method
cpg	273.16	J/molxK	745.56	Joback Method
cpg	280.17	J/molxK	785.21	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	411.50 ± 0.50	K	11.00	NIST Webbook
tbrp	411.70	K	11.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C446106&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C446106&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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