

# 2,4-dichlorobenzyl undecyl ether

<b>Inchi:</b>	InChI=1S/C18H28Cl2O/c1-2-3-4-5-6-7-8-9-10-13-21-15-16-11-12-17(19)14-18(16)20/h1
<b>InchiKey:</b>	OSOLFCOGEAVEPG-UHFFFAOYSA-N
<b>Formula:</b>	C18H28Cl2O
<b>SMILES:</b>	CCCCCCCCCOCCc1ccc(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	331.32

## Physical Properties

Property code	Value	Unit	Source
gf	64.97	kJ/mol	Joback Method
hf	-364.96	kJ/mol	Joback Method
hfus	45.22	kJ/mol	Joback Method
hvap	70.44	kJ/mol	Joback Method
log10ws	-7.41		Crippen Method
logp	7.041		Crippen Method
mcvol	271.070	ml/mol	McGowan Method
pc	1335.88	kPa	Joback Method
rinpol	2287.00		NIST Webbook
rinpol	2287.00		NIST Webbook
rinpol	2275.00		NIST Webbook
rinpol	2281.00		NIST Webbook
rinpol	2284.00		NIST Webbook
rinpol	2284.00		NIST Webbook
rinpol	2274.00		NIST Webbook
rinpol	2280.00		NIST Webbook
rinpol	2284.00		NIST Webbook
rinpol	2287.00		NIST Webbook
rinpol	2275.00		NIST Webbook
rinpol	2281.00		NIST Webbook
rinpol	2284.00		NIST Webbook
rinpol	2280.00		NIST Webbook
rinpol	2275.00		NIST Webbook
tb	745.16	K	Joback Method
tc	940.00	K	Joback Method
tf	426.15	K	Joback Method
vc	1.052	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	749.46	J/molxK	745.16	Joback Method
cpg	766.41	J/molxK	777.63	Joback Method
cpg	782.40	J/molxK	810.11	Joback Method
cpg	797.47	J/molxK	842.58	Joback Method
cpg	811.65	J/molxK	875.06	Joback Method
cpg	824.97	J/molxK	907.53	Joback Method
cpg	837.45	J/molxK	940.00	Joback Method
dvisc	0.0008615	Paxs	426.15	Joback Method
dvisc	0.0004611	Paxs	479.32	Joback Method
dvisc	0.0002797	Paxs	532.49	Joback Method
dvisc	0.0001857	Paxs	585.65	Joback Method
dvisc	0.0001320	Paxs	638.82	Joback Method
dvisc	0.0000989	Paxs	691.99	Joback Method
dvisc	0.0000772	Paxs	745.16	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R32389&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R32389&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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