

# Fendiline

**Other names:**

Benzenepropanamine, «gamma»-phenyl-N-(1-phenylethyl)-  
N-(3,3-Diphenylpropyl)-«alpha»-methylbenzylamine  
N-(3,3-Diphenylpropyl)-1-phenylethylamine  
Benzylamine, N-(3,3-diphenylpropyl)-«alpha»-methyl-  
N-(1-Phenylethyl)-3,3-diphenylpropylamine  
Fendilin  
Phendilin

**Inchi:**

InChI=1S/C23H25N/c1-19(20-11-5-2-6-12-20)24-18-17-23(21-13-7-3-8-14-21)22-15-9-4-

**InchiKey:**

NMKSAYKQLCHXDK-UHFFFAOYSA-N

**Formula:**

C23H25N

**SMILES:**

CC(NCCC(c1ccccc1)c1ccccc1)c1ccccc1

**Mol. weight [g/mol]:**

315.45

**CAS:**

13042-18-7

## Physical Properties

Property code	Value	Unit	Source
gf	564.52	kJ/mol	Joback Method
hf	234.45	kJ/mol	Joback Method
hfus	35.50	kJ/mol	Joback Method
hvap	79.28	kJ/mol	Joback Method
log10ws	-6.47		Crippen Method
logp	5.559		Crippen Method
mcvol	273.630	ml/mol	McGowan Method
pc	1730.34	kPa	Joback Method
rinpol	2420.00		NIST Webbook
tb	854.97	K	Joback Method
tc	1101.99	K	Joback Method
tf	450.89	K	Joback Method
vc	1.022	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	831.91	J/molxK	854.97	Joback Method

cpg	849.71	J/mol×K	896.14	Joback Method
cpg	866.01	J/mol×K	937.31	Joback Method
cpg	880.98	J/mol×K	978.48	Joback Method
cpg	894.74	J/mol×K	1019.65	Joback Method
cpg	907.46	J/mol×K	1060.82	Joback Method
cpg	919.27	J/mol×K	1101.99	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13042187&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13042187&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/36-508-8/Fendiline.pdf>

Generated by Cheméo on 2024-04-26 14:36:15.65553386 +0000 UTC m=+16431424.576111175.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.