

# Ethane-1,2-diyl diisobutyl dicarbonate

Inchi:

InchiKey:

Formula:

SMILES:

Mol. weight [g/mol]:

InChI=1S/C12H22O6/c1-9(2)7-17-11(13)15-5-6-16-12(14)18-8-10(3)4/h9-10H,5-8H2,1-4

HIMCQASYNVYLHG-UHFFFAOYSA-N

C12H22O6

CC(C)COC(=O)OCCOC(=O)OCC(C)C

262.30

## Physical Properties

| Property code | Value    | Unit    | Source         |
|---------------|----------|---------|----------------|
| gf            | -632.56  | kJ/mol  | Joback Method  |
| hf            | -1055.61 | kJ/mol  | Joback Method  |
| hfus          | 27.74    | kJ/mol  | Joback Method  |
| hvap          | 64.66    | kJ/mol  | Joback Method  |
| log10ws       | -2.22    |         | Crippen Method |
| logp          | 2.605    |         | Crippen Method |
| mcvol         | 206.560  | ml/mol  | McGowan Method |
| pc            | 1896.95  | kPa     | Joback Method  |
| rinpol        | 1616.00  |         | NIST Webbook   |
| tb            | 670.50   | K       | Joback Method  |
| tc            | 852.67   | K       | Joback Method  |
| tf            | 383.78   | K       | Joback Method  |
| vc            | 0.779    | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 571.78    | J/mol×K | 670.50          | Joback Method |
| cpg           | 638.28    | J/mol×K | 822.31          | Joback Method |
| cpg           | 626.44    | J/mol×K | 791.94          | Joback Method |
| cpg           | 613.86    | J/mol×K | 761.58          | Joback Method |
| cpg           | 600.54    | J/mol×K | 731.22          | Joback Method |
| cpg           | 586.51    | J/mol×K | 700.86          | Joback Method |
| cpg           | 649.34    | J/mol×K | 852.67          | Joback Method |
| dvisc         | 0.0000786 | Pa×s    | 670.50          | Joback Method |
| dvisc         | 0.0001046 | Pa×s    | 622.71          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0001460 | Pa×s | 574.93 | Joback Method |
| dvisc | 0.0002165 | Pa×s | 527.14 | Joback Method |
| dvisc | 0.0003471 | Pa×s | 479.35 | Joback Method |
| dvisc | 0.0006179 | Pa×s | 431.57 | Joback Method |
| dvisc | 0.0012699 | Pa×s | 383.78 | Joback Method |

## Sources

|                        |                                                                                                                                           |
|------------------------|-------------------------------------------------------------------------------------------------------------------------------------------|
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U378280&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U378280&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                 |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                         |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                     |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                     |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpol:</b>  | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

Latest version available from:

<https://www.chemeo.com/cid/36-541-1/Ethane-1-2-diyl-diisobutyl-dicarbonate.pdf>

Generated by Cheméo on 2025-12-23 06:14:18.00350986 +0000 UTC m=+6218655.533550527.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.