

# Pyrene, 1-methyl-

<b>Other names:</b>	1-Methylpyrene 3-Methylpyrene Pyrene, 3-methyl
<b>Inchi:</b>	InChI=1S/C17H12/c1-11-5-6-14-8-7-12-3-2-4-13-9-10-15(11)17(14)16(12)13/h2-10H,1H3
<b>InchiKey:</b>	KBSPJIWZDWBDGM-UHFFFAOYSA-N
<b>Formula:</b>	C17H12
<b>SMILES:</b>	Cc1ccc2ccc3cccc4ccc1c2c34
<b>Mol. weight [g/mol]:</b>	216.28
<b>CAS:</b>	2381-21-7

## Physical Properties

Property code	Value	Unit	Source
gf	489.97	kJ/mol	Joback Method
hf	335.66	kJ/mol	Joback Method
hfus	26.70	kJ/mol	Joback Method
hvap	92.30 ± 1.30	kJ/mol	NIST Webbook
log10ws	-6.78		Crippen Method
logp	4.892		Crippen Method
mcvol	172.550	ml/mol	McGowan Method
pc	2726.86	kPa	Joback Method
rinpol	370.29		NIST Webbook
rinpol	374.20		NIST Webbook
rinpol	374.43		NIST Webbook
rinpol	373.23		NIST Webbook
rinpol	373.27		NIST Webbook
rinpol	373.44		NIST Webbook
rinpol	373.55		NIST Webbook
rinpol	374.30		NIST Webbook
rinpol	373.72		NIST Webbook
rinpol	373.45		NIST Webbook
rinpol	373.55		NIST Webbook
rinpol	373.40		NIST Webbook
rinpol	373.55		NIST Webbook
rinpol	373.45		NIST Webbook
rinpol	373.55		NIST Webbook
rinpol	373.72		NIST Webbook
rinpol	370.00		NIST Webbook

rinpol	415.27		NIST Webbook
rinpol	373.60		NIST Webbook
rinpol	374.20		NIST Webbook
rinpol	371.86		NIST Webbook
rinpol	372.57		NIST Webbook
rinpol	373.55		NIST Webbook
rinpol	368.09		NIST Webbook
rinpol	370.88		NIST Webbook
rinpol	371.43		NIST Webbook
rinpol	371.73		NIST Webbook
rinpol	373.00		NIST Webbook
rinpol	373.55		NIST Webbook
rinpol	368.09		NIST Webbook
rinpol	370.65		NIST Webbook
rinpol	372.88		NIST Webbook
rinpol	370.29		NIST Webbook
rinpol	373.40		NIST Webbook
rinpol	370.00		NIST Webbook
rinpol	372.97		NIST Webbook
rinpol	371.73		NIST Webbook
rinpol	2220.00		NIST Webbook
rinpol	373.55		NIST Webbook
rinpol	373.44		NIST Webbook
rinpol	2212.00		NIST Webbook
rinpol	2215.00		NIST Webbook
rinpol	2212.00		NIST Webbook
rinpol	372.88		NIST Webbook
rinpol	372.97		NIST Webbook
rinpol	373.05		NIST Webbook
tb	679.22	K	Joback Method
tc	930.34	K	Joback Method
tf	347.65 ± 2.00	K	NIST Webbook
vc	0.675	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	435.75	J/mol×K	679.22	Joback Method
cpg	449.48	J/mol×K	721.07	Joback Method
cpg	462.19	J/mol×K	762.93	Joback Method
cpg	474.06	J/mol×K	804.78	Joback Method

cpg	485.29	J/mol×K	846.63	Joback Method
cpg	496.08	J/mol×K	888.49	Joback Method
cpg	506.62	J/mol×K	930.34	Joback Method
dvisc	0.0019946	Paxs	449.71	Joback Method
dvisc	0.0018069	Paxs	487.96	Joback Method
dvisc	0.0016605	Paxs	526.21	Joback Method
dvisc	0.0015435	Paxs	564.47	Joback Method
dvisc	0.0014482	Paxs	602.72	Joback Method
dvisc	0.0013691	Paxs	640.97	Joback Method
dvisc	0.0013026	Paxs	679.22	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.37781e+01
Coeff. B	-5.05699e+03
Coeff. C	-1.31066e+02
Temperature range (K), min.	505.92
Temperature range (K), max.	728.35

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2381217&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2381217&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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