

# 2,6-dimethyl-2-heptenal

<b>Inchi:</b>	InChI=1S/C9H16O/c1-8(2)5-4-6-9(3)7-10/h6-8H,4-5H2,1-3H3/b9-6+
<b>InchiKey:</b>	YXVDCUSVRUNXEM-RMKNXTFCSA-N
<b>Formula:</b>	C9H16O
<b>SMILES:</b>	CC(C=O)=CCCC(C)C
<b>Mol. weight [g/mol]:</b>	140.22
<b>CAS:</b>	---

## Physical Properties

Property code	Value	Unit	Source
gf	-5.39	kJ/mol	Joback Method
hf	-212.52	kJ/mol	Joback Method
hfus	16.72	kJ/mol	Joback Method
hvap	42.00	kJ/mol	Joback Method
log10ws	-2.48		Crippen Method
logp	2.568		Crippen Method
mcvol	134.940	ml/mol	McGowan Method
pc	2648.83	kPa	Joback Method
rinpol	1020.00		NIST Webbook
tb	457.58	K	Joback Method
tc	642.53	K	Joback Method
tf	199.15	K	Joback Method
vc	0.531	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	282.58	J/mol×K	457.58	Joback Method
cpg	296.03	J/mol×K	488.40	Joback Method
cpg	308.83	J/mol×K	519.23	Joback Method
cpg	321.02	J/mol×K	550.05	Joback Method
cpg	332.61	J/mol×K	580.88	Joback Method
cpg	343.63	J/mol×K	611.70	Joback Method
cpg	354.11	J/mol×K	642.53	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.32166e+01
Coeff. B	-3.54204e+03
Coeff. C	-6.79600e+01
Temperature range (K), min.	341.92
Temperature range (K), max.	516.03

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R145327&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R145327&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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