

1-Decene, 8-methyl-

Other names:	8-Methyl-1-decene 8-Methyldec-1-ene
Inchi:	InChI=1S/C11H22/c1-4-6-7-8-9-10-11(3)5-2/h4,11H,1,5-10H2,2-3H3
InchiKey:	RFSMYVSHKACWAB-UHFFFAOYSA-N
Formula:	C11H22
SMILES:	C=CCCCCCC(C)CC
Mol. weight [g/mol]:	154.29
CAS:	61142-79-8

Physical Properties

Property code	Value	Unit	Source
gf	127.14	kJ/mol	Joback Method
hf	-150.22	kJ/mol	Joback Method
hfus	19.44	kJ/mol	Joback Method
hvap	39.02	kJ/mol	Joback Method
log10ws	-4.04		Crippen Method
logp	4.169		Crippen Method
mcvol	161.550	ml/mol	McGowan Method
pc	2021.76	kPa	Joback Method
rinpol	1061.00		NIST Webbook
rinpol	1062.00		NIST Webbook
rinpol	1056.00		NIST Webbook
rinpol	1017.00		NIST Webbook
tb	447.32	K	Joback Method
tc	615.61	K	Joback Method
tf	196.97	K	Joback Method
vc	0.626	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	343.11	J/mol×K	447.32	Joback Method
cpg	359.03	J/mol×K	475.37	Joback Method
cpg	374.31	J/mol×K	503.42	Joback Method

cpg	388.98	J/mol×K	531.47	Joback Method
cpg	403.04	J/mol×K	559.51	Joback Method
cpg	416.52	J/mol×K	587.56	Joback Method
cpg	429.43	J/mol×K	615.61	Joback Method
dvisc	0.0091557	Paxs	196.97	Joback Method
dvisc	0.0028404	Paxs	238.69	Joback Method
dvisc	0.0012483	Paxs	280.42	Joback Method
dvisc	0.0006788	Paxs	322.14	Joback Method
dvisc	0.0004245	Paxs	363.87	Joback Method
dvisc	0.0002924	Paxs	405.60	Joback Method
dvisc	0.0002159	Paxs	447.32	Joback Method

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.57336e+01
Coeff. B	-4.34242e+03
Coeff. C	-7.01840e+01
Temperature range (K), min.	351.32
Temperature range (K), max.	486.84

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C61142798&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity

gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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