

# Dipropyl sulfoxide

<b>Other names:</b>	(n-C <sub>3</sub> H <sub>7</sub> ) <sub>2</sub> SO 1,1'-Sulphinylbispropane 1,1'-sulfinylbispropane Di-n-propyl sulfoxid Di-n-propyl sulphoxide Di-n-propylsulfoxide Dipropyl sulphoxide NSC 75838 Propane, 1,1'-sulfinylbis- Sulfoxide, dipropyl di-n-propyl sulfoxide propyl sulfoxide
<b>Inchi:</b>	InChI=1S/C6H14OS/c1-3-5-8(7)6-4-2/h3-6H2,1-2H3
<b>InchiKey:</b>	BQCCJWMQESHLIT-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>6</sub> H <sub>14</sub> OS
<b>SMILES:</b>	CCCS(=O)CCC
<b>Mol. weight [g/mol]:</b>	134.24
<b>CAS:</b>	4253-91-2

## Physical Properties

Property code	Value	Unit	Source
chl	-4634.60 ± 0.40	kJ/mol	NIST Webbook
gf	-218.07	kJ/mol	Joback Method
hf	-255.00 ± 2.00	kJ/mol	NIST Webbook
hfl	-329.00 ± 0.80	kJ/mol	NIST Webbook
hfus	19.05	kJ/mol	Joback Method
hvap	74.00 ± 1.00	kJ/mol	NIST Webbook
ie	8.60	eV	NIST Webbook
log10ws	-0.97		Crippen Method
logp	1.555		Crippen Method
mcvol	117.620	ml/mol	McGowan Method
pc	3427.87	kPa	Joback Method
tb	475.00 ± 1.00	K	NIST Webbook
tc	570.30	K	Joback Method
tf	297.90 ± 1.00	K	NIST Webbook
vc	0.462	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	217.98	J/molxK	394.96	Joback Method
cpg	229.40	J/molxK	424.18	Joback Method
cpg	240.43	J/molxK	453.41	Joback Method
cpg	251.06	J/molxK	482.63	Joback Method
cpg	261.30	J/molxK	511.85	Joback Method
cpg	271.16	J/molxK	541.08	Joback Method
cpg	280.63	J/molxK	570.30	Joback Method
srf	0.03	N/m	308.15	Surface and bulk behavior of (dialkylsulfoxides + carbon tetrachloride) mixtures
srf	0.03	N/m	303.15	Surface and bulk behavior of (dialkylsulfoxides + carbon tetrachloride) mixtures
srf	0.03	N/m	298.15	Surface and bulk behavior of (dialkylsulfoxides + carbon tetrachloride) mixtures
srf	0.03	N/m	313.15	Surface and bulk behavior of (dialkylsulfoxides + carbon tetrachloride) mixtures
srf	0.03	N/m	298.15	Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures
srf	0.03	N/m	303.15	Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures

srf	0.03	N/m	308.15	Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures
srf	0.03	N/m	313.15	Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures
srf	0.03	N/m	318.15	Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures
srf	0.03	N/m	323.15	Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures
srf	0.03	N/m	328.15	Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4253912&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4253912&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Surface and bulk behavior of (dialkylsulfoxides + carbon tetrachloride) and Surface Tension and Refractive Index of Dialkylsulfoxide + Water Mixtures at Several Temperatures:</b>	<a href="https://www.doi.org/10.1016/j.jct.2009.06.021">https://www.doi.org/10.1016/j.jct.2009.06.021</a>
	<a href="https://www.doi.org/10.1021/je7001013">https://www.doi.org/10.1021/je7001013</a>
	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

chl: Standard liquid enthalpy of combustion

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>srf:</b>	Surface Tension
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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