

«beta»-Alanine, N-(4-trifluoromethylbenzoyl)-, hexadecyl ester

Inchi: InChI=1S/C27H42F3NO3/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-22-34-25(32)20-21-31-2
InchiKey: XXWMNUOYDQEQSB-UHFFFAOYSA-N
Formula: C27H42F3NO3
SMILES: CCCCCCCCCCCCCCCCCOC(=O)CCNC(=O)c1ccc(C(F)(F)F)cc1
Mol. weight [g/mol]: 485.62

Physical Properties

Property code	Value	Unit	Source
gf	-575.80	kJ/mol	Joback Method
hf	-1276.54	kJ/mol	Joback Method
hfus	70.65	kJ/mol	Joback Method
hvap	97.22	kJ/mol	Joback Method
log10ws	-9.32		Crippen Method
logp	7.850		Crippen Method
mcvol	391.830	ml/mol	McGowan Method
pc	821.95	kPa	Joback Method
tb	1023.73	K	Joback Method
tc	1262.40	K	Joback Method
tf	611.93	K	Joback Method
vc	1.548	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1358.38	J/mol×K	1023.73	Joback Method
cpg	1376.51	J/mol×K	1063.51	Joback Method
cpg	1393.19	J/mol×K	1103.29	Joback Method
cpg	1408.57	J/mol×K	1143.07	Joback Method
cpg	1422.76	J/mol×K	1182.84	Joback Method
cpg	1435.89	J/mol×K	1222.62	Joback Method
cpg	1448.09	J/mol×K	1262.40	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321749&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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