

# N,N-Dimethyl-N'-butyl-p-methylbenzamidine

Inchi:	InChI=1S/C14H22N2/c1-5-6-11-15-14(16(3)4)13-9-7-12(2)8-10-13/h7-10H,5-6,11H2,1-4H
InchiKey:	CVNOZHFXCZQWEO-CCEZHUSRSA-N
Formula:	C14H22N2
SMILES:	CCCCN=C(c1ccc(C)cc1)N(C)C
Mol. weight [g/mol]:	218.34

## Physical Properties

Property code	Value	Unit	Source
hf	32.73	kJ/mol	Joback Method
hvap	55.13	kJ/mol	Joback Method
log10ws	-3.17		Crippen Method
logp	3.103		Crippen Method
mcvol	200.020	ml/mol	McGowan Method
pc	1813.86	kPa	Joback Method
rinpol	1627.00		NIST Webbook
rinpol	1627.00		NIST Webbook
tb	640.38	K	Joback Method
tc	851.18	K	Joback Method

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R159123&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R159123&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

hf:	Enthalpy of formation at standard conditions
hvap:	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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