

# Thiazolidine, 2-propyl-

<b>Other names:</b>	2-n-Propylthiazolidine 2-Propyl-1,3-thiazolidine 2-Propylthiazolidine
<b>Inchi:</b>	InChI=1S/C6H13NS/c1-2-3-6-7-4-5-8-6/h6-7H,2-5H2,1H3
<b>InchiKey:</b>	UDJJSTNTIKWCCW-UHFFFAOYSA-N
<b>Formula:</b>	C6H13NS
<b>SMILES:</b>	CCCC1NCCS1
<b>Mol. weight [g/mol]:</b>	131.24
<b>CAS:</b>	24050-10-0

## Physical Properties

Property code	Value	Unit	Source
gf	163.76	kJ/mol	Joback Method
hf	-23.62	kJ/mol	Joback Method
hfus	18.48	kJ/mol	Joback Method
hvap	41.78	kJ/mol	Joback Method
log10ws	-1.91		Crippen Method
logp	1.449		Crippen Method
mcvol	110.870	ml/mol	McGowan Method
pc	3930.78	kPa	Joback Method
rinpol	1135.00		NIST Webbook
rinpol	1081.00		NIST Webbook
rinpol	1086.00		NIST Webbook
rinpol	1136.00		NIST Webbook
ripol	1615.00		NIST Webbook
ripol	1615.00		NIST Webbook
ripol	1610.00		NIST Webbook
tb	448.34	K	Joback Method
tc	669.43	K	Joback Method
tf	356.76	K	Joback Method
vc	0.396	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	221.26	J/mol×K	448.34	Joback Method
cpg	235.77	J/mol×K	485.19	Joback Method
cpg	249.52	J/mol×K	522.04	Joback Method
cpg	262.53	J/mol×K	558.88	Joback Method
cpg	274.84	J/mol×K	595.73	Joback Method
cpg	286.46	J/mol×K	632.58	Joback Method
cpg	297.42	J/mol×K	669.43	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24050100&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24050100&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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