

# Benactyzine

<b>Other names:</b>	Benzeneacetic acid, «alpha»-hydroxy-«alpha»-phenyl-, 2-(diethylamino)ethyl ester Benzilic acid, 2-(diethylamino)ethyl ester Benactyzin Benactizina Benzilic acid «beta»-diethylaminoethyl ester Diazil Diethylaminoethyl benzilate «beta»-Diethylaminoethyl benzilate 2-(Diethylamino)ethyl benzilate 2-(Diethylamino)ethyl diphenylglycolate Diphenylglycolic acid 2-(diethylamino)ethyl ester «alpha»-Hydroxy-«alpha»-phenylbenzeneacetic acid 2-(diethylamino)ethyl ester 2-Diethylaminoethyl benizilate Diethyl(2-hydroxyethyl)amine benzilate Benactizine
<b>Inchi:</b>	InChI=1S/C20H25NO3/c1-3-21(4-2)15-16-24-19(22)20(23,17-11-7-5-8-12-17)18-13-9-6-
<b>InchiKey:</b>	IVQOFBKHQCTVQV-UHFFFAOYSA-N
<b>Formula:</b>	C20H25NO3
<b>SMILES:</b>	CCN(CC)CCOC(=O)C(O)(c1ccccc1)c1ccccc1
<b>Mol. weight [g/mol]:</b>	327.42
<b>CAS:</b>	302-40-9

## Physical Properties

Property code	Value	Unit	Source
gf	85.22	kJ/mol	Joback Method
hf	-321.32	kJ/mol	Joback Method
hfus	38.12	kJ/mol	Joback Method
hvap	91.25	kJ/mol	Joback Method
log10ws	-3.37		Crippen Method
logp	2.808		Crippen Method
mcvol	268.430	ml/mol	McGowan Method
pc	1856.31	kPa	Joback Method
rinpol	2248.00		NIST Webbook
rinpol	2249.00		NIST Webbook
rinpol	2235.00		NIST Webbook
rinpol	2248.00		NIST Webbook
rinpol	2248.00		NIST Webbook

rinpol	2250.00		NIST Webbook
rinpol	2285.00		NIST Webbook
rinpol	2235.00		NIST Webbook
rinpol	2249.00		NIST Webbook
rinpol	2230.00		NIST Webbook
tb	888.04	K	Joback Method
tc	1105.29	K	Joback Method
tf	535.87	K	Joback Method
vc	0.990	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	840.21	J/mol×K	888.04	Joback Method
cpg	854.04	J/mol×K	924.25	Joback Method
cpg	866.85	J/mol×K	960.46	Joback Method
cpg	878.73	J/mol×K	996.67	Joback Method
cpg	889.79	J/mol×K	1032.88	Joback Method
cpg	900.12	J/mol×K	1069.08	Joback Method
cpg	909.80	J/mol×K	1105.29	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C302409&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C302409&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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