

Benactyzine

Other names:	Benzeneacetic acid, «alpha»-hydroxy-«alpha»-phenyl-, 2-(diethylamino)ethyl ester Benzilic acid, 2-(diethylamino)ethyl ester Benactyzin Benactizina Benzilic acid «beta»-diethylaminoethyl ester Diazil Diethylaminoethyl benzilate «beta»-Diethylaminoethyl benzilate 2-(Diethylamino)ethyl benzilate 2-(Diethylamino)ethyl diphenylglycolate Diphenylglycolic acid 2-(diethylamino)ethyl ester «alpha»-Hydroxy-«alpha»-phenylbenzeneacetic acid 2-(diethylamino)ethyl ester 2-Diethylaminoethyl benizilate Diethyl(2-hydroxyethyl)amine benzilate Benactizine
Inchi:	InChI=1S/C20H25NO3/c1-3-21(4-2)15-16-24-19(22)20(23,17-11-7-5-8-12-17)18-13-9-6-
InchiKey:	IVQOFBKHQCTVQV-UHFFFAOYSA-N
Formula:	C20H25NO3
SMILES:	CCN(CC)CCOC(=O)C(O)(c1ccccc1)c1ccccc1
Mol. weight [g/mol]:	327.42
CAS:	302-40-9

Physical Properties

Property code	Value	Unit	Source
gf	85.22	kJ/mol	Joback Method
hf	-321.32	kJ/mol	Joback Method
hfus	38.12	kJ/mol	Joback Method
hvap	91.25	kJ/mol	Joback Method
log10ws	-3.37		Crippen Method
logp	2.808		Crippen Method
mcvol	268.430	ml/mol	McGowan Method
pc	1856.31	kPa	Joback Method
rinpol	2235.00		NIST Webbook
rinpol	2249.00		NIST Webbook
rinpol	2249.00		NIST Webbook
rinpol	2248.00		NIST Webbook
rinpol	2230.00		NIST Webbook

rinp	2248.00		NIST Webbook
rinp	2250.00		NIST Webbook
rinp	2285.00		NIST Webbook
rinp	2235.00		NIST Webbook
rinp	2248.00		NIST Webbook
tb	888.04	K	Joback Method
tc	1105.29	K	Joback Method
tf	535.87	K	Joback Method
vc	0.990	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	840.21	J/mol×K	888.04	Joback Method
cpg	854.04	J/mol×K	924.25	Joback Method
cpg	866.85	J/mol×K	960.46	Joback Method
cpg	878.73	J/mol×K	996.67	Joback Method
cpg	889.79	J/mol×K	1032.88	Joback Method
cpg	900.12	J/mol×K	1069.08	Joback Method
cpg	909.80	J/mol×K	1105.29	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C302409&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
mc_{vol}:	McGowan's characteristic volume
p_c:	Critical Pressure
r_{inpol}:	Non-polar retention indices
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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