

# Cyclopentanecarboxylic acid, 2-oxo-, ethyl ester

<b>Other names:</b>	«alpha»-(Carboethoxy)cyclopentanone Ethyl cyclopentanone-2-carboxylate Ethyl 2-cyclopentanone-1-carboxylate Ethyl 2-cyclopentanonecarboxylate Ethyl 2-oxo-1-cyclopentanecarboxylate Ethyl 2-oxocyclopentanecarboxylate Ethyl 2-oxocyclopentanecarboxylate 2-(Ethoxycarbonyl)cyclopentanone 2-Carboethoxy-1-cyclopentanone 2-Carboethoxycyclopentanone 2-Cyclopentanonecarboxylic acid ethyl ester Cyclopentanone-2-carboxylic acid ethyl ester 2-Carboethoxycyclopentanone Ethyl ester of 2-oxocyclopentanecarboxylic acid 2-(Ethoxycarbonyl)-1-cyclopentanone
<b>Inchi:</b>	InChI=1S/C8H12O3/c1-2-11-8(10)6-4-3-5-7(6)9/h6H,2-5H2,1H3
<b>InchiKey:</b>	JHZPNBKZPAWCJD-UHFFFAOYSA-N
<b>Formula:</b>	C8H12O3
<b>SMILES:</b>	CCOC(=O)C1CCCC1=O
<b>Mol. weight [g/mol]:</b>	156.18
<b>CAS:</b>	611-10-9

## Physical Properties

Property code	Value	Unit	Source
gf	-303.48	kJ/mol	Joback Method
hf	-530.47	kJ/mol	Joback Method
hfus	12.71	kJ/mol	Joback Method
hvap	47.06	kJ/mol	Joback Method
log10ws	-0.97		Crippen Method
logp	0.919		Crippen Method
mcvol	121.730	ml/mol	McGowan Method
pc	3360.64	kPa	Joback Method
tb	541.83	K	Joback Method
tc	761.25	K	Joback Method
tf	331.20	K	Joback Method
vc	0.456	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	292.08	J/mol×K	541.83	Joback Method
cpg	306.74	J/mol×K	578.40	Joback Method
cpg	320.71	J/mol×K	614.97	Joback Method
cpg	333.97	J/mol×K	651.54	Joback Method
cpg	346.52	J/mol×K	688.11	Joback Method
cpg	358.33	J/mol×K	724.68	Joback Method
cpg	369.38	J/mol×K	761.25	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	376.20	K	1.50	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C611109&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C611109&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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