

# Propanoic acid, 2-chloro, 4-pentenyl ester

<b>Inchi:</b>	InChI=1S/C8H13ClO2/c1-3-4-5-6-11-8(10)7(2)9/h3,7H,1,4-6H2,2H3
<b>InchiKey:</b>	YQFYHTOIOXUBMM-UHFFFAOYSA-N
<b>Formula:</b>	C8H13ClO2
<b>SMILES:</b>	C=CCCCOC(=O)C(C)Cl
<b>Mol. weight [g/mol]:</b>	176.64

## Physical Properties

Property code	Value	Unit	Source
gf	-143.97	kJ/mol	Joback Method
hf	-348.84	kJ/mol	Joback Method
hfus	18.66	kJ/mol	Joback Method
hvap	45.89	kJ/mol	Joback Method
log10ws	-2.15		Crippen Method
logp	2.123		Crippen Method
mcvol	138.960	ml/mol	McGowan Method
pc	2693.00	kPa	Joback Method
rinpol	1106.00		NIST Webbook
rinpol	1101.00		NIST Webbook
rinpol	1123.00		NIST Webbook
rinpol	1122.00		NIST Webbook
rinpol	1128.00		NIST Webbook
ripol	1546.00		NIST Webbook
ripol	1552.00		NIST Webbook
ripol	1554.00		NIST Webbook
tb	492.40	K	Joback Method
tc	680.78	K	Joback Method
tf	265.24	K	Joback Method
vc	0.531	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	290.22	J/mol×K	492.40	Joback Method
cpg	301.76	J/mol×K	523.80	Joback Method

cpg	312.79	J/mol×K	555.19	Joback Method
cpg	323.33	J/mol×K	586.59	Joback Method
cpg	333.38	J/mol×K	617.98	Joback Method
cpg	342.95	J/mol×K	649.38	Joback Method
cpg	352.05	J/mol×K	680.78	Joback Method
dvisc	0.0039024	Paxs	265.24	Joback Method
dvisc	0.0018579	Paxs	303.10	Joback Method
dvisc	0.0010430	Paxs	340.96	Joback Method
dvisc	0.0006571	Paxs	378.82	Joback Method
dvisc	0.0004503	Paxs	416.68	Joback Method
dvisc	0.0003286	Paxs	454.54	Joback Method
dvisc	0.0002517	Paxs	492.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R113781&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R113781&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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