

# 1H-Indene-1,2-diol, 2,3-dihydro-, cis-

<b>Other names:</b>	1,2-Indandiol, cis- cis-1,2-Indandiol Indan, 1,2-dihydroxy-, cis- cis-1,2-Indanediol cis-Indan-1,2-diol
<b>Inchi:</b>	InChI=1S/C9H10O2/c10-8-5-6-3-1-2-4-7(6)9(8)11/h1-4,8-11H,5H2/t8-,9+/m1/s1
<b>InchiKey:</b>	YKXXBEOXRPZVCC-BDAKNGLRSA-N
<b>Formula:</b>	C9H10O2
<b>SMILES:</b>	OC1Cc2ccccc2C1O
<b>Mol. weight [g/mol]:</b>	150.17
<b>CAS:</b>	4647-42-1

## Physical Properties

Property code	Value	Unit	Source
chs	-4592.00 ± 4.00	kJ/mol	NIST Webbook
gf	-92.92	kJ/mol	Joback Method
hf	-256.03	kJ/mol	Joback Method
hfus	20.10	kJ/mol	Joback Method
hvap	71.53	kJ/mol	Joback Method
log10ws	-1.65		Crippen Method
logp	0.637		Crippen Method
mcvol	114.790	ml/mol	McGowan Method
pc	4534.68	kPa	Joback Method
rinpol	1400.00		NIST Webbook
tb	623.41	K	Joback Method
tc	820.36	K	Joback Method
tf	365.47	K	Joback Method
vc	0.425	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	301.01	J/mol×K	623.41	Joback Method
cpg	310.98	J/mol×K	656.23	Joback Method

cpg	320.34	J/molxK	689.06	Joback Method
cpg	329.11	J/molxK	721.88	Joback Method
cpg	337.35	J/molxK	754.71	Joback Method
cpg	345.09	J/molxK	787.53	Joback Method
cpg	352.37	J/molxK	820.36	Joback Method
dvisc	0.0058361	Paxs	365.47	Joback Method
dvisc	0.0019449	Paxs	408.46	Joback Method
dvisc	0.0007990	Paxs	451.45	Joback Method
dvisc	0.0003832	Paxs	494.44	Joback Method
dvisc	0.0002067	Paxs	537.43	Joback Method
dvisc	0.0001222	Paxs	580.42	Joback Method
dvisc	0.0000776	Paxs	623.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4647421&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4647421&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

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