

# Dimethylmalonic acid, 1-phenyl-2-(cyclohex-2-enyl)ethyl tridecyl

InChI:  
ester

InChI=1S/C32H50O4/c1-4-5-6-7-8-9-10-11-12-13-20-25-35-30(33)32(2,3)31(34)36-29(28)

InChIKey:

NTAGOVGMGYPSMP-UHFFFAOYSA-N

Formula:

C32H50O4

SMILES:

CCCCCCCCCCCCOC(=O)C(C)(C)C(=O)OC(CC1C=CCCC1)c1ccccc1

Mol. weight [g/mol]:

498.74

## Physical Properties

Property code	Value	Unit	Source
gf	-82.06	kJ/mol	Joback Method
hf	-858.81	kJ/mol	Joback Method
hfus	60.37	kJ/mol	Joback Method
hvap	106.45	kJ/mol	Joback Method
log10ws	-9.77		Crippen Method
logp	8.898		Crippen Method
mvol	437.700	ml/mol	McGowan Method
pc	772.03	kPa	Joback Method
rinpol	3294.00		NIST Webbook
rinpol	3294.00		NIST Webbook
tb	1125.86	K	Joback Method
tc	1382.96	K	Joback Method
tf	616.70	K	Joback Method
vc	1.669	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1584.81	J/molxK	1125.86	Joback Method
cpg	1601.89	J/molxK	1168.71	Joback Method
cpg	1617.14	J/molxK	1211.56	Joback Method
cpg	1630.75	J/molxK	1254.41	Joback Method
cpg	1642.86	J/molxK	1297.26	Joback Method
cpg	1653.66	J/molxK	1340.11	Joback Method
cpg	1663.30	J/molxK	1382.96	Joback Method
dvisc	0.0001630	Paxs	616.70	Joback Method

dvisc	0.0000688	Paxs	701.56	Joback Method
dvisc	0.0000350	Paxs	786.42	Joback Method
dvisc	0.0000203	Paxs	871.28	Joback Method
dvisc	0.0000129	Paxs	956.14	Joback Method
dvisc	0.0000089	Paxs	1041.00	Joback Method
dvisc	0.0000065	Paxs	1125.86	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U361879&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U361879&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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