

# Propane, 2-[(1,1-dimethylethyl)sulfonyl]-2-methyl-

Other names:	Di-tert-Butyl sulfone tert-Butyl Sulfone Di-tert-butyl sulphone 2-((1,1-Dimethylethyl)-sulphonyl)-2-methylpropane
Inchi:	InChI=1S/C8H18O2S/c1-7(2,3)11(9,10)8(4,5)6/h1-6H3
InchiKey:	SPJQDMKTFSPPLO-UHFFFAOYSA-N
Formula:	C8H18O2S
SMILES:	CC(C)(C)S(=O)(=O)C(C)(C)C
Mol. weight [g/mol]:	178.29
CAS:	1886-75-5

## Physical Properties

Property code	Value	Unit	Source
chs	-5682.50 ± 2.20	kJ/mol	NIST Webbook
gf	-446.38	kJ/mol	Joback Method
hf	-546.30 ± 2.50	kJ/mol	NIST Webbook
hfs	-640.40 ± 2.30	kJ/mol	NIST Webbook
hfus	13.03	kJ/mol	Joback Method
hvap	49.45	kJ/mol	Joback Method
log10ws	-2.23		Crippen Method
logp	1.998		Crippen Method
mcvol	151.670	ml/mol	McGowan Method
pc	3110.57	kPa	Joback Method
tb	521.00 ± 3.00	K	NIST Webbook
tc	606.70	K	Joback Method
tf	403.00 ± 2.00	K	NIST Webbook
tf	403.00 ± 3.00	K	NIST Webbook
vc	0.588	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	313.21	J/mol×K	423.76	Joback Method
cpg	329.64	J/mol×K	454.25	Joback Method

cpg	345.21	J/mol×K	484.74	Joback Method
cpg	359.95	J/mol×K	515.23	Joback Method
cpg	373.87	J/mol×K	545.72	Joback Method
cpg	387.01	J/mol×K	576.21	Joback Method
cpg	399.40	J/mol×K	606.70	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1886755&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1886755&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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