

# 1-Naphthoic acid, 2,3,4,6-tetrachlorophenyl ester

Inchi:	InChI=1S/C17H8Cl4O2/c18-12-8-13(19)16(15(21)14(12)20)23-17(22)11-7-3-5-9-4-1-2-6
InchiKey:	OHNLBGHBSDGQKI-UHFFFAOYSA-N
Formula:	C17H8Cl4O2
SMILES:	O=C(Oc1c(Cl)cc(Cl)c(Cl)c1Cl)c1cccc2ccccc12
Mol. weight [g/mol]:	386.06

## Physical Properties

Property code	Value	Unit	Source
gf	93.94	kJ/mol	Joback Method
hf	-95.19	kJ/mol	Joback Method
hfus	42.52	kJ/mol	Joback Method
hvap	89.63	kJ/mol	Joback Method
log10ws	-8.10		Crippen Method
logp	6.673		Crippen Method
mcvol	239.810	ml/mol	McGowan Method
pc	2224.99	kPa	Joback Method
rinpol	3002.00		NIST Webbook
tb	911.61	K	Joback Method
tc	1177.04	K	Joback Method
tf	621.33	K	Joback Method
vc	0.913	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	576.78	J/molxK	911.61	Joback Method
cpg	585.81	J/molxK	955.85	Joback Method
cpg	593.96	J/molxK	1000.09	Joback Method
cpg	601.30	J/molxK	1044.32	Joback Method
cpg	607.93	J/molxK	1088.56	Joback Method
cpg	613.94	J/molxK	1132.80	Joback Method
cpg	619.42	J/molxK	1177.04	Joback Method
dvisc	0.0005156	Paxs	621.33	Joback Method
dvisc	0.0003865	Paxs	669.71	Joback Method

dvisc	0.0003012	Paxs	718.09	Joback Method
dvisc	0.0002423	Paxs	766.47	Joback Method
dvisc	0.0002000	Paxs	814.85	Joback Method
dvisc	0.0001686	Paxs	863.23	Joback Method
dvisc	0.0001448	Paxs	911.61	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U355696&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U355696&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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