

# trans-(2,3,3a,7a-Tetrahydro-1H-indene-4-carbalde

Inchi:	InChI=1S/C10H12O/c11-7-9-5-1-3-8-4-2-6-10(8)9/h1,3,5,7-8,10H,2,4,6H2/t8-,10-/m0/s1
InchiKey:	NRPGDWVRKOJMEE-WPRPVWTQSA-N
Formula:	C10H12O
SMILES:	O=CC1=CC=CC2CCCC12
Mol. weight [g/mol]:	148.20

## Physical Properties

Property code	Value	Unit	Source
gf	69.29	kJ/mol	Joback Method
hf	-104.10	kJ/mol	Joback Method
hfus	15.97	kJ/mol	Joback Method
hvap	46.16	kJ/mol	Joback Method
log10ws	-2.30		Crippen Method
logp	2.098		Crippen Method
mcvol	123.010	ml/mol	McGowan Method
pc	3411.87	kPa	Joback Method
rinpol	1254.00		NIST Webbook
rinpol	1254.00		NIST Webbook
ripol	1848.00		NIST Webbook
ripol	1848.00		NIST Webbook
tb	506.45	K	Joback Method
tc	730.39	K	Joback Method
tf	283.82	K	Joback Method
vc	0.474	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	281.32	J/molxK	506.45	Joback Method
cpg	297.58	J/molxK	543.77	Joback Method
cpg	312.72	J/molxK	581.10	Joback Method
cpg	326.82	J/molxK	618.42	Joback Method
cpg	339.93	J/molxK	655.74	Joback Method
cpg	352.11	J/molxK	693.07	Joback Method

cpg	363.43	J/mol×K	730.39	Joback Method
dvisc	0.0021391	Paxs	283.82	Joback Method
dvisc	0.0015377	Paxs	320.93	Joback Method
dvisc	0.0011837	Paxs	358.03	Joback Method
dvisc	0.0009570	Paxs	395.13	Joback Method
dvisc	0.0008026	Paxs	432.24	Joback Method
dvisc	0.0006920	Paxs	469.34	Joback Method
dvisc	0.0006098	Paxs	506.45	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R586753&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R586753&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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