

# Cyclohexane, hexadecyl-

<b>Other names:</b>	n-Hexadecyl-cyclohexane
<b>Inchi:</b>	InChI=1S/C22H44/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-16-19-22-20-17-15-18-21-22/h22H
<b>InchiKey:</b>	NRHBFNBZAZZTAU-UHFFFAOYSA-N
<b>Formula:</b>	C22H44
<b>SMILES:</b>	CCCCCCCCCCCCCCCC1CCCCC1
<b>Mol. weight [g/mol]:</b>	308.58
<b>CAS:</b>	6812-38-0

## Physical Properties

Property code	Value	Unit	Source
chl	-14485.30 ± 3.50	kJ/mol	NIST Webbook
gf	158.81	kJ/mol	Joback Method
hf	-460.50 ± 3.70	kJ/mol	NIST Webbook
hfus	44.57	kJ/mol	Joback Method
hvap	109.30	kJ/mol	NIST Webbook
log10ws	-8.68		Crippen Method
logp	8.438		Crippen Method
mcvol	309.980	ml/mol	McGowan Method
pc	1024.00	kPa	Joback Method
rinpol	2285.70		NIST Webbook
rinpol	2279.00		NIST Webbook
rinpol	2280.10		NIST Webbook
rinpol	2272.00		NIST Webbook
ripol	2304.20		NIST Webbook
tb	722.31	K	Joback Method
tc	900.92	K	Joback Method
tf	308.80 ± 1.00	K	NIST Webbook
tf	305.70 ± 1.00	K	NIST Webbook
tf	304.15 ± 2.00	K	NIST Webbook
tf	305.20 ± 2.00	K	NIST Webbook
tf	306.90 ± 2.00	K	NIST Webbook
vc	1.200	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	961.67	J/molxK	722.31	Joback Method
cpg	985.23	J/molxK	752.08	Joback Method
cpg	1007.60	J/molxK	781.85	Joback Method
cpg	1028.83	J/molxK	811.61	Joback Method
cpg	1048.97	J/molxK	841.38	Joback Method
cpg	1068.04	J/molxK	871.15	Joback Method
cpg	1086.10	J/molxK	900.92	Joback Method
dvisc	0.0035167	Paxs	345.08	Joback Method
dvisc	0.0011482	Paxs	407.95	Joback Method
dvisc	0.0005055	Paxs	470.82	Joback Method
dvisc	0.0002700	Paxs	533.69	Joback Method
dvisc	0.0001646	Paxs	596.57	Joback Method
dvisc	0.0001103	Paxs	659.44	Joback Method
dvisc	0.0000792	Paxs	722.31	Joback Method
hvapt	79.60	kJ/mol	598.00	NIST Webbook

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6812380&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6812380&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/37-129-8/Cyclohexane-hexadecyl.pdf>

Generated by Cheméo on 2023-03-26 02:37:28.377666551 +0000 UTC m=+843616.272790570.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.