

# Resorcinol, 2-acetyl-

<b>Other names:</b>	2',6'-Dihydroxyacetophenone 2,6-Dihydroxyacetophenone 2-Acetylresorcinol Acetyl-2,6-dihydroxybenzene Ethanone, 1-(2,6-dihydroxyphenyl)- «gamma»-Resacetophenone Acetophenone, 2',6'-dihydroxy- 1,3-Benzenediol, 2-acetyl- 1-(2,6-Dihydroxyphenyl)ethanone NSC 615
<b>Inchi:</b>	InChI=1S/C8H8O3/c1-5(9)8-6(10)3-2-4-7(8)11/h2-4,10-11H,1H3
<b>InchiKey:</b>	YPTJKHVBDCKNF-UHFFFAOYSA-N
<b>Formula:</b>	C8H8O3
<b>SMILES:</b>	CC(=O)c1c(O)cccc1O
<b>Mol. weight [g/mol]:</b>	152.15
<b>CAS:</b>	699-83-2

## Physical Properties

Property code	Value	Unit	Source
gf	-309.27	kJ/mol	Joback Method
hf	-439.12	kJ/mol	Joback Method
hfus	23.68	kJ/mol	Joback Method
hvap	68.45	kJ/mol	Joback Method
log10ws	-1.23		Crippen Method
logp	1.300		Crippen Method
mcvol	113.130	ml/mol	McGowan Method
pc	5972.16	kPa	Joback Method
tb	624.23	K	Joback Method
tc	869.83	K	Joback Method
tf	479.71	K	Joback Method
vc	0.314	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.53	J/mol×K	624.23	Joback Method
cpg	317.34	J/mol×K	828.90	Joback Method
cpg	310.35	J/mol×K	787.96	Joback Method
cpg	303.12	J/mol×K	747.03	Joback Method
cpg	295.51	J/mol×K	706.10	Joback Method
cpg	287.36	J/mol×K	665.16	Joback Method
cpg	324.25	J/mol×K	869.83	Joback Method
dvisc	0.0000059	Paxs	624.23	Joback Method
dvisc	0.0000089	Paxs	600.14	Joback Method
dvisc	0.0000138	Paxs	576.06	Joback Method
dvisc	0.0000224	Paxs	551.97	Joback Method
dvisc	0.0000380	Paxs	527.88	Joback Method
dvisc	0.0000676	Paxs	503.80	Joback Method
dvisc	0.0001275	Paxs	479.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C699832&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C699832&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/37-216-1/Resorcinol-2-acetyl.pdf>

Generated by Cheméo on 2024-04-20 07:45:25.410599774 +0000 UTC m=+15888374.331177089.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.