

# Propane, 1,1,1,2-tetrachloro-

<b>Other names:</b>	1,1,1,2-Tetrachloropropane CH <sub>3</sub> CHClCCl <sub>3</sub>
<b>Inchi:</b>	InChI=1S/C3H4Cl4/c1-2(4)3(5,6)7/h2H,1H3
<b>InchiKey:</b>	FEKGWIHDBVDVSM-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>3</sub> H <sub>4</sub> Cl <sub>4</sub>
<b>SMILES:</b>	CC(Cl)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	181.88
<b>CAS:</b>	812-03-3

## Physical Properties

Property code	Value	Unit	Source
gf	-72.94	kJ/mol	Joback Method
hf	-182.24	kJ/mol	Joback Method
hfus	9.38	kJ/mol	Joback Method
hvap	38.13	kJ/mol	Joback Method
log10ws	-2.90		Crippen Method
logp	2.984		Crippen Method
mvol	102.090	ml/mol	McGowan Method
pc	3777.68	kPa	Joback Method
rinpol	903.00		NIST Webbook
rinpol	918.00		NIST Webbook
tb	414.09	K	Joback Method
tc	633.01	K	Joback Method
tf	209.00 ± 4.00	K	NIST Webbook
vc	0.383	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	152.43	J/mol×K	414.09	Joback Method
cpg	158.97	J/mol×K	450.58	Joback Method
cpg	164.94	J/mol×K	487.06	Joback Method
cpg	170.36	J/mol×K	523.55	Joback Method
cpg	175.28	J/mol×K	560.04	Joback Method

cpg	179.73	J/mol×K	596.53	Joback Method
cpg	183.76	J/mol×K	633.01	Joback Method
dvisc	0.0088117	Paxs	230.67	Joback Method
dvisc	0.0040006	Paxs	261.24	Joback Method
dvisc	0.0021431	Paxs	291.81	Joback Method
dvisc	0.0012923	Paxs	322.38	Joback Method
dvisc	0.0008506	Paxs	352.95	Joback Method
dvisc	0.0005985	Paxs	383.52	Joback Method
dvisc	0.0004436	Paxs	414.09	Joback Method
hvapt	42.30	kJ/mol	400.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.26725e+01
Coeff. B	-3.05911e+03
Coeff. C	-6.28950e+01
Temperature range (K), min.	309.90
Temperature range (K), max.	478.48

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C812033&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C812033&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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