

# 3-Vinyl-1,2-dithiacyclohex-4-ene

<b>Other names:</b>	1,2-Dithi-4-ene, 3-ethenyl 3-Ethenyl-1,2-dithi-4-ene 3-Vinyl-1,2-dithi-4-ene 3-Vinyl-1,2-dithiocylohex-4-ene
<b>Inchi:</b>	InChI=1S/C6H8S2/c1-2-6-4-3-5-7-8-6/h2-4,6H,1,5H2
<b>InchiKey:</b>	UQXHSMWBRGWFBK-UHFFFAOYSA-N
<b>Formula:</b>	C6H8S2
<b>SMILES:</b>	C=CC1C=CCSS1
<b>Mol. weight [g/mol]:</b>	144.26
<b>CAS:</b>	62488-52-2

## Physical Properties

Property code	Value	Unit	Source
gf	221.61	kJ/mol	Joback Method
hf	160.88	kJ/mol	Joback Method
hfus	10.39	kJ/mol	Joback Method
hvap	40.62	kJ/mol	Joback Method
log10ws	-2.81		Crippen Method
logp	2.492		Crippen Method
mcvol	108.640	ml/mol	McGowan Method
pc	4311.22	kPa	Joback Method
rinpol	1205.00		NIST Webbook
rinpol	1180.00		NIST Webbook
rinpol	1214.00		NIST Webbook
rinpol	1190.90		NIST Webbook
rinpol	1205.00		NIST Webbook
ripol	1750.00		NIST Webbook
ripol	1750.00		NIST Webbook
tb	447.73	K	Joback Method
tc	695.41	K	Joback Method
tf	330.66	K	Joback Method
vc	0.363	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	194.06	J/mol×K	447.73	Joback Method
cpg	207.13	J/mol×K	489.01	Joback Method
cpg	219.27	J/mol×K	530.29	Joback Method
cpg	230.54	J/mol×K	571.57	Joback Method
cpg	240.97	J/mol×K	612.85	Joback Method
cpg	250.62	J/mol×K	654.13	Joback Method
cpg	259.53	J/mol×K	695.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C62488522&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C62488522&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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