

Pentaerithrol, tetrakis-(2-ethylhexanoate)

Other names:	2,2-bis[[[2-ethyl-1-oxohexyl]oxy]methyl]propane-1,3-diyl bis(2-ethylhexanoate) Pentaerythritol tetra-2-ethylhexanoate pentaerythritol tetra(2-ethylhexanoate)
Inchi:	InChI=1S/C37H68O8/c1-9-17-21-29(13-5)33(38)42-25-37(26-43-34(39)30(14-6)22-18-10
InchiKey:	DRRMRHKHTQRWMB-UHFFFAOYSA-N
Formula:	C37H68O8
SMILES:	CCCCC(CC)C(=O)OCC(COC(=O)C(CC)CCCC)(COC(=O)C(CC)CCCC)COC(=O)C(CC)
Mol. weight [g/mol]:	640.93
CAS:	7299-99-2

Physical Properties

Property code	Value	Unit	Source
gf	-681.94	kJ/mol	Joback Method
hf	-1816.08	kJ/mol	Joback Method
hfus	81.23	kJ/mol	Joback Method
hvap	131.73	kJ/mol	Joback Method
log10ws	-9.55		Crippen Method
logp	9.011		Crippen Method
mcvol	561.950	ml/mol	McGowan Method
pc	486.23	kPa	Joback Method
tb	1346.13	K	Joback Method
tc	1851.81	K	Joback Method
tf	737.81	K	Joback Method
vc	2.168	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	2103.58	J/molxK	1683.25	Joback Method
cpg	2123.46	J/molxK	1514.69	Joback Method
cpg	2122.16	J/molxK	1430.41	Joback Method
cpg	2112.21	J/molxK	1346.13	Joback Method
cpg	2084.16	J/molxK	1767.53	Joback Method
cpg	2059.58	J/molxK	1851.81	Joback Method

cpg	2116.97	J/molxK	1598.97	Joback Method
dvisc	0.0000090	Paxs	839.20	Joback Method
dvisc	0.0000044	Paxs	940.58	Joback Method
dvisc	0.0000024	Paxs	1041.97	Joback Method
dvisc	0.0000015	Paxs	1143.36	Joback Method
dvisc	0.0000010	Paxs	1244.74	Joback Method
dvisc	0.0000007	Paxs	1346.13	Joback Method
dvisc	0.0000228	Paxs	737.81	Joback Method
hvapt	126.40	kJ/mol	399.00	NIST Webbook
pvap	6.44e-04	kPa	443.17	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	2.87e-04	kPa	433.18	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	1.33e-04	kPa	423.40	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	5.82e-05	kPa	413.44	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	2.37e-05	kPa	403.44	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants

pvap	8.94e-06	kPa	393.56	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	3.37e-06	kPa	383.56	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	1.07e-06	kPa	373.81	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	3.67e-07	kPa	363.89	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants
pvap	1.27e-07	kPa	353.96	Vapor pressure measurements in the range 10 ⁻⁵ Pa to 1 Pa of four pentaerythritol esters Density and vapor liquid equilibria modeling of ester lubricants

Datasets

Mass density, kg/m³

Temperature, K - Liquid	Pressure, kPa - Liquid	Mass density, kg/m3 - Liquid
278.15	100.00	971.86
278.15	1000.00	972.35
278.15	5000.00	974.57
278.15	10000.00	977.26
278.15	15000.00	979.87
278.15	20000.00	982.42
278.15	25000.00	984.85
278.15	30000.00	987.27
278.15	35000.00	989.6
278.15	40000.00	991.89
278.15	45000.00	994.12
283.15	100.00	968.45
283.15	1000.00	968.97
283.15	5000.00	971.24
283.15	10000.00	974.05
283.15	15000.00	976.74
283.15	20000.00	979.35
283.15	25000.00	981.87
283.15	30000.00	984.34
283.15	35000.00	986.75
283.15	40000.00	989.06
283.15	45000.00	991.35
293.15	100.00	961.62
293.15	1000.00	962.16
293.15	5000.00	964.55
293.15	10000.00	967.46
293.15	15000.00	970.3
293.15	20000.00	973.0
293.15	25000.00	975.67
293.15	30000.00	978.27
293.15	35000.00	980.77
293.15	40000.00	983.23
293.15	45000.00	985.61
303.15	100.00	954.72
303.15	1000.00	955.27
303.15	5000.00	957.79
303.15	10000.00	960.83
303.15	15000.00	963.79
303.15	20000.00	966.65
303.15	25000.00	969.41
303.15	30000.00	972.11
303.15	35000.00	974.73
303.15	40000.00	977.27

303.15	45000.00	979.76
313.15	100.00	947.88
313.15	1000.00	948.35
313.15	5000.00	950.98
313.15	10000.00	954.14
313.15	15000.00	957.23
313.15	20000.00	960.21
313.15	25000.00	963.1
313.15	30000.00	965.89
313.15	35000.00	968.62
313.15	40000.00	971.26
313.15	45000.00	973.82
323.15	100.00	940.45
323.15	1000.00	941.25
323.15	5000.00	943.86
323.15	10000.00	947.2
323.15	15000.00	950.44
323.15	20000.00	953.54
323.15	25000.00	956.57
323.15	30000.00	959.48
323.15	35000.00	962.32
323.15	40000.00	965.0
323.15	45000.00	967.75
333.15	100.00	933.58
333.15	1000.00	934.21
333.15	5000.00	937.13
333.15	10000.00	940.66
333.15	15000.00	944.02
333.15	20000.00	947.27
333.15	25000.00	950.4
333.15	30000.00	953.41
333.15	35000.00	956.37
333.15	40000.00	959.21
333.15	45000.00	961.98
343.15	100.00	926.49
343.15	1000.00	927.17
343.15	5000.00	930.23
343.15	10000.00	933.89
343.15	15000.00	937.42
343.15	20000.00	940.81
343.15	25000.00	944.09
343.15	30000.00	947.23
343.15	35000.00	950.27
343.15	40000.00	953.23

343.15	45000.00	956.09
353.15	100.00	919.45
353.15	1000.00	920.14
353.15	5000.00	923.34
353.15	10000.00	927.13
353.15	15000.00	930.84
353.15	20000.00	934.37
353.15	25000.00	937.77
353.15	30000.00	941.05
353.15	35000.00	944.19
353.15	40000.00	947.27
353.15	45000.00	950.24

Reference

<https://www.doi.org/10.1021/je049580w>

Sources

Compressed Liquid Densities of Squalane and Pentaerythritol Tetra-2-ethylhexanoate):

<https://www.doi.org/10.1021/je049580w>

Joback Method:

https://www.chemeo.com/doc/models/crippen_log10ws

McGowan Method:

https://en.wikipedia.org/wiki/Joback_method

Solubility of Carbon Dioxide in Two Pentaerythritol Ester Oils between (283 K and 303 K):

<http://link.springer.com/article/10.1007/BF02311772>

Vapor pressure measurements in the range 10⁻⁵ Pa to 1 Pa of four pentaerythritol esters under pressure for mixtures of dimethyl ether and ester hydrocarbons: Density-Viscosity Relationship:

<https://www.doi.org/10.1021/je800178x>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C7299992&Units=SI>

<https://www.doi.org/10.1016/j.fluid.2007.07.029>

<https://www.doi.org/10.1021/je700121n>

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume

pc:	Critical Pressure
pvap:	Vapor pressure
rho:	Liquid Density
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/37-443-9/Pentaerithrol-tetrakis-2-ethylhexanoate.pdf>

Generated by Cheméo on 2024-04-25 17:15:20.978603688 +0000 UTC m=+16354569.899181003.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.