

# Trifluoroacetic acid,n-tridecyl ester

**Other names:**

Tridecyl trifluoroacetate  
Tridecyl 2,2,2-trifluoroacetate  
1-Tridecanol, trifluoroacetate  
Trifluoroacetic acid, tridecyl ester

**Inchi:**

InChI=1S/C15H27F3O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-20-14(19)15(16,17)18/h2-13H2,

**InchiKey:**

DKWJQENYIDPCEX-UHFFFAOYSA-N

**Formula:**

C15H27F3O2

**SMILES:**

CCCCCCCCCCCCCOC(=O)C(F)(F)F

**Mol. weight [g/mol]:**

296.37

**CAS:**

53800-02-5

## Physical Properties

| Property code | Value    | Unit    | Source         |
|---------------|----------|---------|----------------|
| gf            | -740.09  | kJ/mol  | Joback Method  |
| hf            | -1194.81 | kJ/mol  | Joback Method  |
| hfus          | 39.22    | kJ/mol  | Joback Method  |
| hvap          | 54.39    | kJ/mol  | Joback Method  |
| log10ws       | -5.63    |         | Crippen Method |
| logp          | 5.403    |         | Crippen Method |
| mcvol         | 234.960  | ml/mol  | McGowan Method |
| pc            | 1343.73  | kPa     | Joback Method  |
| rinpol        | 1525.00  |         | NIST Webbook   |
| rinpol        | 1527.50  |         | NIST Webbook   |
| rinpol        | 1525.00  |         | NIST Webbook   |
| tb            | 613.47   | K       | Joback Method  |
| tc            | 771.41   | K       | Joback Method  |
| tf            | 335.16   | K       | Joback Method  |
| vc            | 0.943    | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value  | Unit    | Temperature [K] | Source        |
|---------------|--------|---------|-----------------|---------------|
| cpg           | 647.26 | J/molxK | 613.47          | Joback Method |
| cpg           | 663.52 | J/molxK | 639.79          | Joback Method |

|     |        |         |        |               |
|-----|--------|---------|--------|---------------|
| cpg | 679.08 | J/mol×K | 666.12 | Joback Method |
| cpg | 693.94 | J/mol×K | 692.44 | Joback Method |
| cpg | 708.13 | J/mol×K | 718.76 | Joback Method |
| cpg | 721.67 | J/mol×K | 745.09 | Joback Method |
| cpg | 734.58 | J/mol×K | 771.41 | Joback Method |

## Sources

|                        |   |
|------------------------|---|
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>   |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C53800025&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C53800025&amp;Units=SI</a> |
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |

## Legend

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mcvol:</b>   | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>rinpola:</b> | Non-polar retention indices                     |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

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