

# Pyridine, 2,5-dimethyl-

<b>Other names:</b>	2,5-Dimethylpyridine 2,5-LUTIDINE 5-methyl-2-methylpyridine
<b>Inchi:</b>	InChI=1S/C7H9N/c1-6-3-4-7(2)8-5-6/h3-5H,1-2H3
<b>InchiKey:</b>	XWKFPIDWVPLX-UHFFFAOYSA-N
<b>Formula:</b>	C7H9N
<b>SMILES:</b>	<chem>Cc1ccc(C)nc1</chem>
<b>Mol. weight [g/mol]:</b>	107.15
<b>CAS:</b>	589-93-5

## Physical Properties

Property code	Value	Unit	Source
affp	958.80	kJ/mol	NIST Webbook
basg	926.90	kJ/mol	NIST Webbook
chl	-4059.40	kJ/mol	NIST Webbook
dm	2.20	debye	KDB
hf	66.44	kJ/mol	KDB
hf	66.40	kJ/mol	NIST Webbook
hfl	18.60	kJ/mol	NIST Webbook
hvap	47.82	kJ/mol	NIST Webbook
ie	8.80 ± 0.05	eV	NIST Webbook
log10ws	0.40		Aqueous Solubility Prediction Method
logp	1.698		Crippen Method
mcvol	95.710	ml/mol	McGowan Method
rinpol	926.00		NIST Webbook
rinpol	936.00		NIST Webbook
rinpol	915.00		NIST Webbook
rinpol	913.00		NIST Webbook
rinpol	919.10		NIST Webbook
rinpol	909.20		NIST Webbook
rinpol	928.00		NIST Webbook
rinpol	936.00		NIST Webbook
rinpol	903.20		NIST Webbook
rinpol	928.00		NIST Webbook
rinpol	916.00		NIST Webbook
rinpol	916.00		NIST Webbook

rinpol	909.20		NIST Webbook
rinpol	915.00		NIST Webbook
rinpol	919.00		NIST Webbook
rinpol	930.00		NIST Webbook
rinpol	912.00		NIST Webbook
rinpol	919.10		NIST Webbook
rinpol	920.00		NIST Webbook
rinpol	921.00		NIST Webbook
rinpol	917.00		NIST Webbook
rinpol	912.00		NIST Webbook
rinpol	926.00		NIST Webbook
rinpol	913.00		NIST Webbook
rinpol	911.00		NIST Webbook
rinpol	917.00		NIST Webbook
rinpol	912.00		NIST Webbook
rinpol	913.00		NIST Webbook
rinpol	946.00		NIST Webbook
rinpol	926.00		NIST Webbook
ripol	1395.00		NIST Webbook
ripol	1371.00		NIST Webbook
ripol	1329.00		NIST Webbook
ripol	1325.00		NIST Webbook
ripol	1371.00		NIST Webbook
ripol	1326.00		NIST Webbook
ripol	1412.00		NIST Webbook
ripol	1325.00		NIST Webbook
ripol	1330.00		NIST Webbook
ripol	1335.00		NIST Webbook
ripol	1332.00		NIST Webbook
ripol	1329.00		NIST Webbook
ripol	1415.00		NIST Webbook
ripol	1370.00		NIST Webbook
ripol	1373.00		NIST Webbook
ripol	1372.00		NIST Webbook
ripol	1330.00		NIST Webbook
sl	248.81	J/molxK	NIST Webbook
tb	430.16	K	KDB
tb	429.90 ± 0.30	K	NIST Webbook
tb	426.25 ± 1.50	K	NIST Webbook
tb	429.15 ± 2.00	K	NIST Webbook
tb	426.65 ± 3.00	K	NIST Webbook
tb	431.20	K	NIST Webbook
tc	644.15 ± 1.00	K	NIST Webbook
tc	644.16 ± 0.30	K	NIST Webbook

tc	644.20	K	KDB
tf	257.61 ± 0.03	K	NIST Webbook
tf	257.00	K	KDB
tf	259.07	K	NIST Webbook
tf	257.15 ± 0.50	K	NIST Webbook
tt	259.07 ± 0.01	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpl	184.70	J/mol×K	298.15	NIST Webbook
hfust	14.64	kJ/mol	259.10	NIST Webbook
hfust	14.64	kJ/mol	259.10	NIST Webbook
hvapt	44.40	kJ/mol	400.50	NIST Webbook
hvapt	41.90	kJ/mol	400.50	NIST Webbook
hvapt	39.40	kJ/mol	400.50	NIST Webbook
hvapt	36.50	kJ/mol	400.50	NIST Webbook
hvapt	42.80	kJ/mol	394.50	NIST Webbook
rho1	938.00	kg/m <sup>3</sup>	273.00	KDB

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.46024e+01
Coeff. B	-3.69877e+03
Coeff. C	-6.07320e+01
Temperature range (K), min.	319.12
Temperature range (K), max.	458.84

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/T + C \cdot \ln(T) + D \cdot T^2$
Coeff. A	9.69764e+01
Coeff. B	-8.86176e+03
Coeff. C	-1.20654e+01

Coeff. D	7.61858e-06
Temperature range (K), min.	350.00
Temperature range (K), max.	435.00

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C589935&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C589935&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>KDB Vapor Pressure Data:</b>	<a href="https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=1356">https://www.chemic.org/research/kdb/hcprop/showprop.php?cmpid=1356</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>KDB:</b>	<a href="https://www.chemic.org/files/research/kdb/mol/mol1356.mol">https://www.chemic.org/files/research/kdb/mol/mol1356.mol</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpl:</b>	Liquid phase heat capacity
<b>dm:</b>	Dipole Moment
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pvap:</b>	Vapor pressure
<b>rho:</b>	Liquid Density
<b>rinp:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**tt:** Triple Point Temperature

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