

# 2-Propanol, 1-chloro-3-isobutoxy

<b>Inchi:</b>	InChI=1S/C7H15ClO2/c1-6(2)4-10-5-7(9)3-8/h6-7,9H,3-5H2,1-2H3
<b>InchiKey:</b>	OYLSNRGLVWHWNI-UHFFFAOYSA-N
<b>Formula:</b>	C7H15ClO2
<b>SMILES:</b>	CC(C)COCC(O)CCl
<b>Mol. weight [g/mol]:</b>	166.65

## Physical Properties

Property code	Value	Unit	Source
gf	-250.57	kJ/mol	Joback Method
hf	-498.56	kJ/mol	Joback Method
hfus	16.31	kJ/mol	Joback Method
hvap	53.87	kJ/mol	Joback Method
log10ws	-1.13		Crippen Method
logp	1.259		Crippen Method
mcvol	133.470	ml/mol	McGowan Method
pc	2966.57	kPa	Joback Method
rinpol	1130.00		NIST Webbook
rinpol	1130.00		NIST Webbook
tb	510.71	K	Joback Method
tc	683.29	K	Joback Method
tf	251.62	K	Joback Method
vc	0.501	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	298.61	J/molxK	510.71	Joback Method
cpg	309.22	J/molxK	539.47	Joback Method
cpg	319.42	J/molxK	568.24	Joback Method
cpg	329.22	J/molxK	597.00	Joback Method
cpg	338.62	J/molxK	625.76	Joback Method
cpg	347.63	J/molxK	654.53	Joback Method
cpg	356.25	J/molxK	683.29	Joback Method
dvisc	0.0578400	Paxs	251.62	Joback Method

dvisc	0.0097512	Paxs	294.80	Joback Method
dvisc	0.0025909	Paxs	337.98	Joback Method
dvisc	0.0009295	Paxs	381.16	Joback Method
dvisc	0.0004109	Paxs	424.35	Joback Method
dvisc	0.0002112	Paxs	467.53	Joback Method
dvisc	0.0001215	Paxs	510.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R313916&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R313916&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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