

# Propanedioic acid, dibromo-, dimethyl ester

<b>Other names:</b>	dimethyl dibromomalonate
<b>Inchi:</b>	InChI=1S/C5H6Br2O4/c1-10-3(8)5(6,7)4(9)11-2/h1-2H3
<b>InchiKey:</b>	UWZGBUTYDAIEAS-UHFFFAOYSA-N
<b>Formula:</b>	C5H6Br2O4
<b>SMILES:</b>	COC(=O)C(Br)(Br)C(=O)OC
<b>Mol. weight [g/mol]:</b>	289.91
<b>CAS:</b>	37167-59-2

## Physical Properties

Property code	Value	Unit	Source
gf	-445.14	kJ/mol	Joback Method
hf	-592.22	kJ/mol	Joback Method
hfus	17.44	kJ/mol	Joback Method
hvap	56.61	kJ/mol	Joback Method
log10ws	-1.11		Crippen Method
logp	0.818		Crippen Method
mcvol	131.190	ml/mol	McGowan Method
pc	4762.81	kPa	Joback Method
rinpol	1252.00		NIST Webbook
tb	595.47	K	Joback Method
tc	827.23	K	Joback Method
tf	412.45	K	Joback Method
vc	0.476	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	258.13	J/molxK	595.47	Joback Method
cpg	265.65	J/molxK	634.10	Joback Method
cpg	272.60	J/molxK	672.72	Joback Method
cpg	279.01	J/molxK	711.35	Joback Method
cpg	284.90	J/molxK	749.98	Joback Method
cpg	290.29	J/molxK	788.60	Joback Method
cpg	295.21	J/molxK	827.23	Joback Method

dvisc	0.0014055	Paxs	412.45	Joback Method
dvisc	0.0009529	Paxs	442.95	Joback Method
dvisc	0.0006792	Paxs	473.46	Joback Method
dvisc	0.0005044	Paxs	503.96	Joback Method
dvisc	0.0003875	Paxs	534.46	Joback Method
dvisc	0.0003063	Paxs	564.97	Joback Method
dvisc	0.0002480	Paxs	595.47	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C37167592&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C37167592&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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