

# Iodoquinol

**Other names:** 5,7-Diiodo-8-hydroxyquinoline  
5,7-Diiodo-8-quinolinol  
5,7-Diiodo-oxine  
5,7-Diiodo-8-quinolinol  
8-Hydroxy-5,7-diiodoquinoline  
8-Quinolinol, 5,7-diiodo-  
Di-Quinol  
Diamoebin  
Diiodohydroxyquin  
Diiodohydroxyquinoline  
Diiodoquin  
Dijodoxichinoline  
Dinoleine  
Diodohydroxyquin  
Diodoquin  
Diodoquine  
Diodoxilin  
Direxide  
Disoquin  
Dyodin  
Embequin  
Enterodiamoebin  
Enterosept  
Floraquin  
Fluoraquin  
Iloquin  
Iloquin suspension  
Lanodoxin  
Meobiquin  
Moebiquin  
NSC 8704  
Quinadome  
Rafamebin  
SS 578  
Searlequin  
Searlequin  
Sebaquin  
Stanquinat  
Yodoxin  
Zoaquin

**Inchi:** InChI=1S/C9H5I2NO/c10-6-4-7(11)9(13)8-5(6)2-1-3-12-8/h1-4,13H  
**InchiKey:** UXZFQZANDVDGMM-UHFFFAOYSA-N  
**Formula:** C9H5I2NO  
**SMILES:** Oc1c(I)cc(I)c2cccnc12  
**Mol. weight [g/mol]:** 396.95  
**CAS:** 83-73-8

## Physical Properties

Property code	Value	Unit	Source
chs	-4349.70 ± 1.20	kJ/mol	NIST Webbook
hf	220.30 ± 1.90	kJ/mol	NIST Webbook
hfs	93.50 ± 1.70	kJ/mol	NIST Webbook
hsub	126.80 ± 0.80	kJ/mol	NIST Webbook
hsub	111.00 ± 0.80	kJ/mol	NIST Webbook
log10ws	-4.89		Crippen Method
logp	3.150		Crippen Method
mcvol	161.940	ml/mol	McGowan Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hsubt	121.90 ± 0.80	kJ/mol	396.50	NIST Webbook
hsubt	110.90	kJ/mol	413.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.53169e+01
Coeff. B	-1.33320e+04
Temperature range (K), min.	532.65
Temperature range (K), max.	666.42

# Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C83738&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C83738&amp;Units=SI</a>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pvap:</b>	Vapor pressure

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